

Unlocking the Potential of Nature's Dendrimer: From Serendipitous Discovery to Fundamental Science to Commercialization

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of **GUELPH**

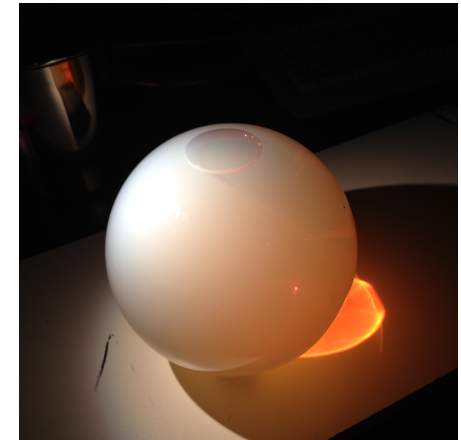


Outline

- natural phytoglycogen nanoparticles
 - unique structure
 - highly branched particles – dendrimer
 - nano-confined water
 - soft & deformable
 - soft colloid & glassy physics
 - PhytoSpherix natural nanotechnology
 - Mirexus
- summary

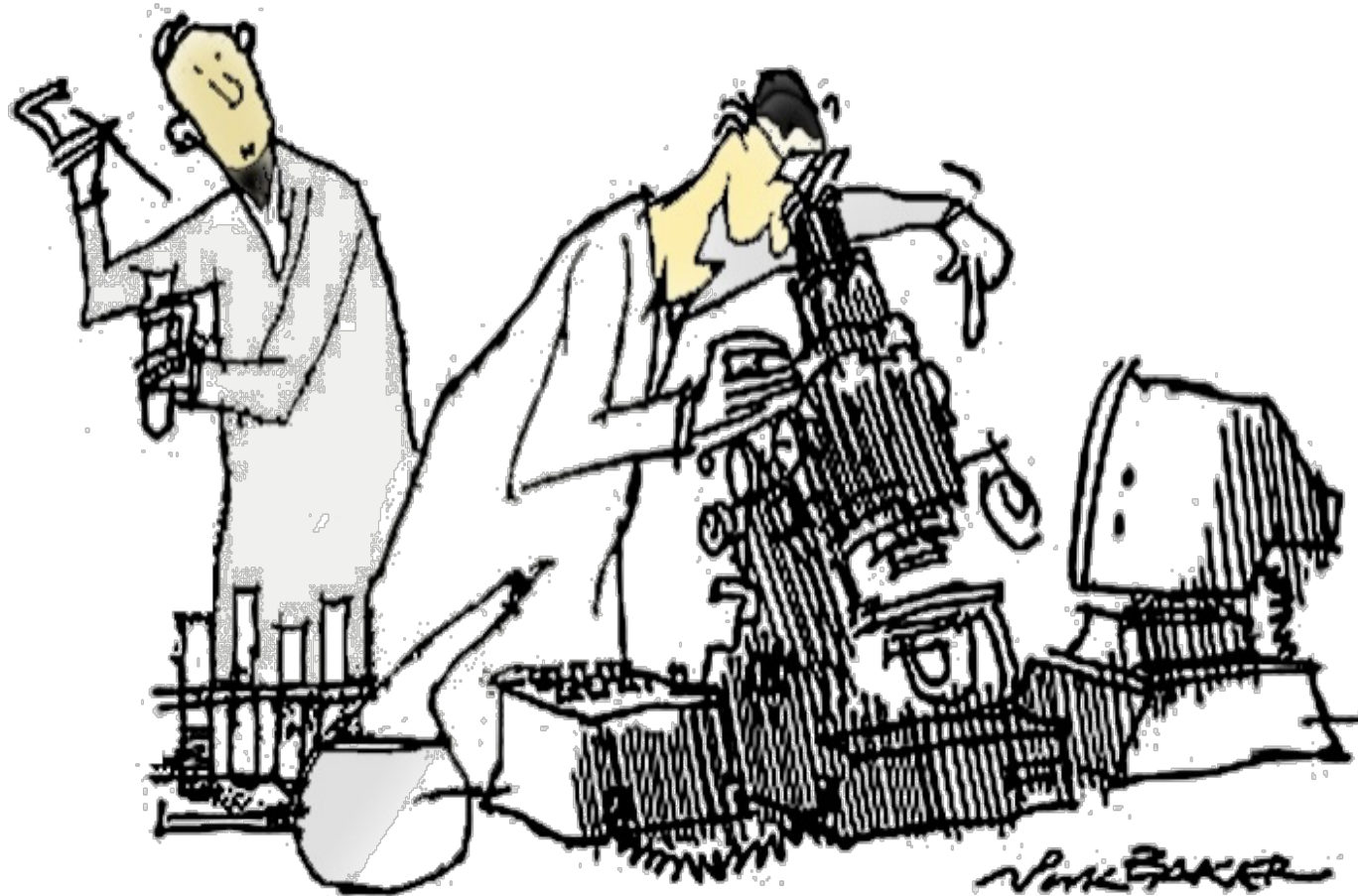
Serendipitous Discovery

- fundamental, multidisciplinary collaboration involving physics & biology
- complicated, multi-step chemical process
- waste product from one step
 - interesting optical properties
 - opalescence
- kept waste instead of throwing it out
- eventually measured on atomic force microscope

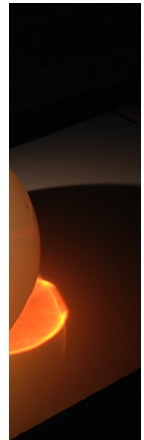


Serendipitous Discovery

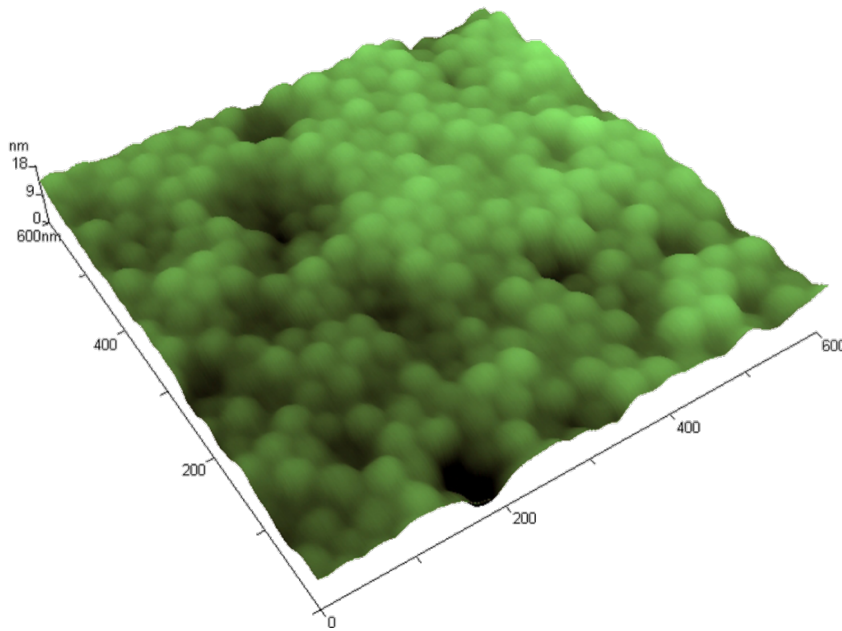
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Phytoglycogen Nanoparticles

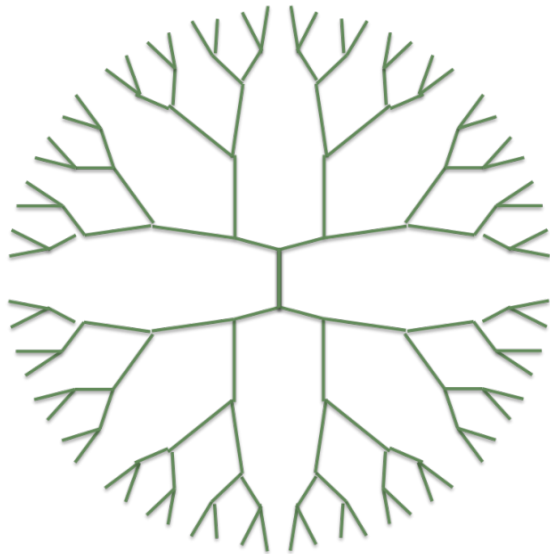


PhytoSpherix™

- uniform nano-size (35 nm dia)
 - highly-branched, soft, compact nanoparticles
- uniform surface chemistry
 - tunable surface properties
- biodegradable & edible
 - extracted & purified in their natural state from sweet corn
 - no harsh chemicals or enzymes

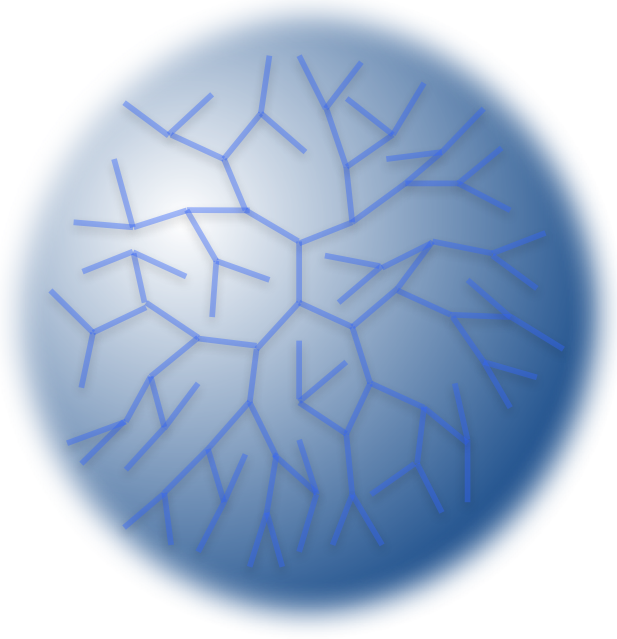


Phytoglycogen Nanoparticles



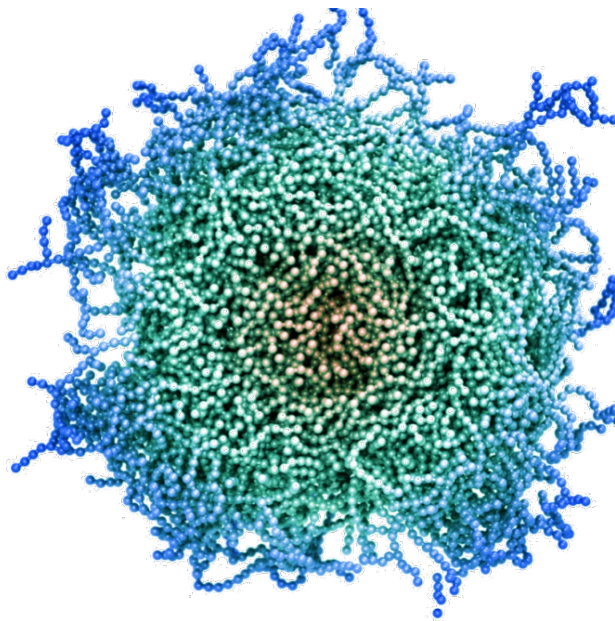
- Nature's dendrimer
 - glucose polymer with dendrimeric structure
- highly-branched spherical single molecule

Phytoglycogen Nanoparticles



- Nature's dendrimer
 - glucose polymer with dendrimeric structure
- highly-branched spherical single molecule
- real structure has
 - uniform density core
 - highly-branched throughout

Phytoglycogen Nanoparticles



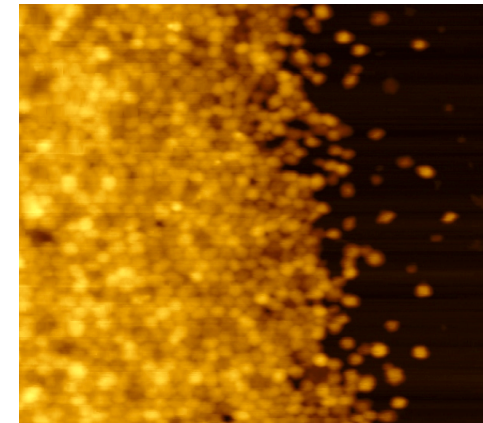
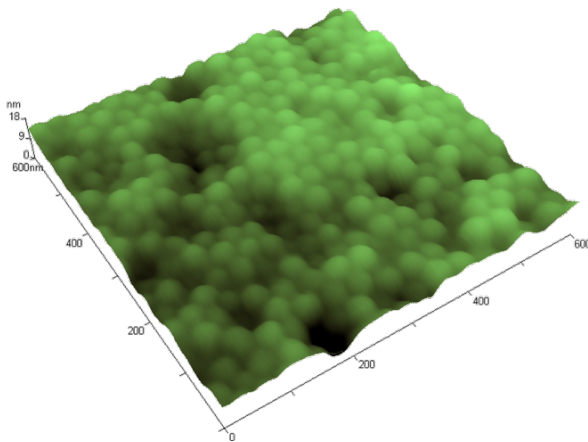
- Nature's dendrimer
 - glucose polymer with dendrimeric structure
- highly-branched spherical single molecule
- real structure has
 - uniform density core
 - highly-branched throughout

[de Haan (2017)]

Unique Properties of Phytoglycogen

- special structure results in special properties
 - monodispersity
 - extraordinary water retention
 - low viscosity in water
 - high stability in water
 - ability to stabilize & disperse bioactive compounds
 - film forming capacity

} special interaction with water



Techniques To Study Water

- neutron scattering

- scattering of neutrons from particles dispersed in water
 - uniform particle size & density, molar mass
 - high water content & slow water dynamics

[Nickels *et al.*, *Biomacromolecules* (2016)]

- infrared spectroscopy

- absorption of infrared by O-H bonds
 - highly structured water inside particles

[Grossutti & Dutcher, *Biomacromolecules* (2016)]

- ellipsometry

- swelling of ultrathin films of particles

- short-range hydration forces [Grossutti *et al.*, *Langmuir* (2017)]

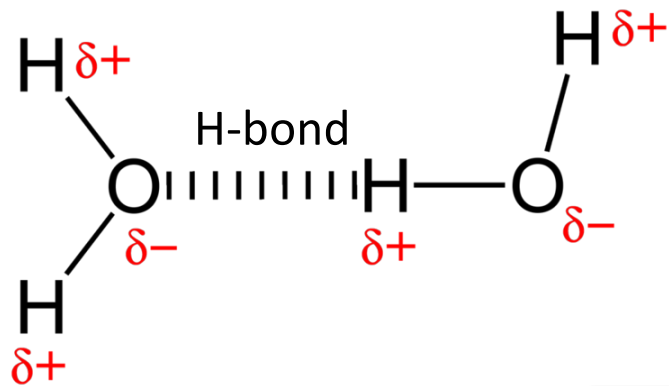
- rheology

- viscosity of aqueous dispersions of particles

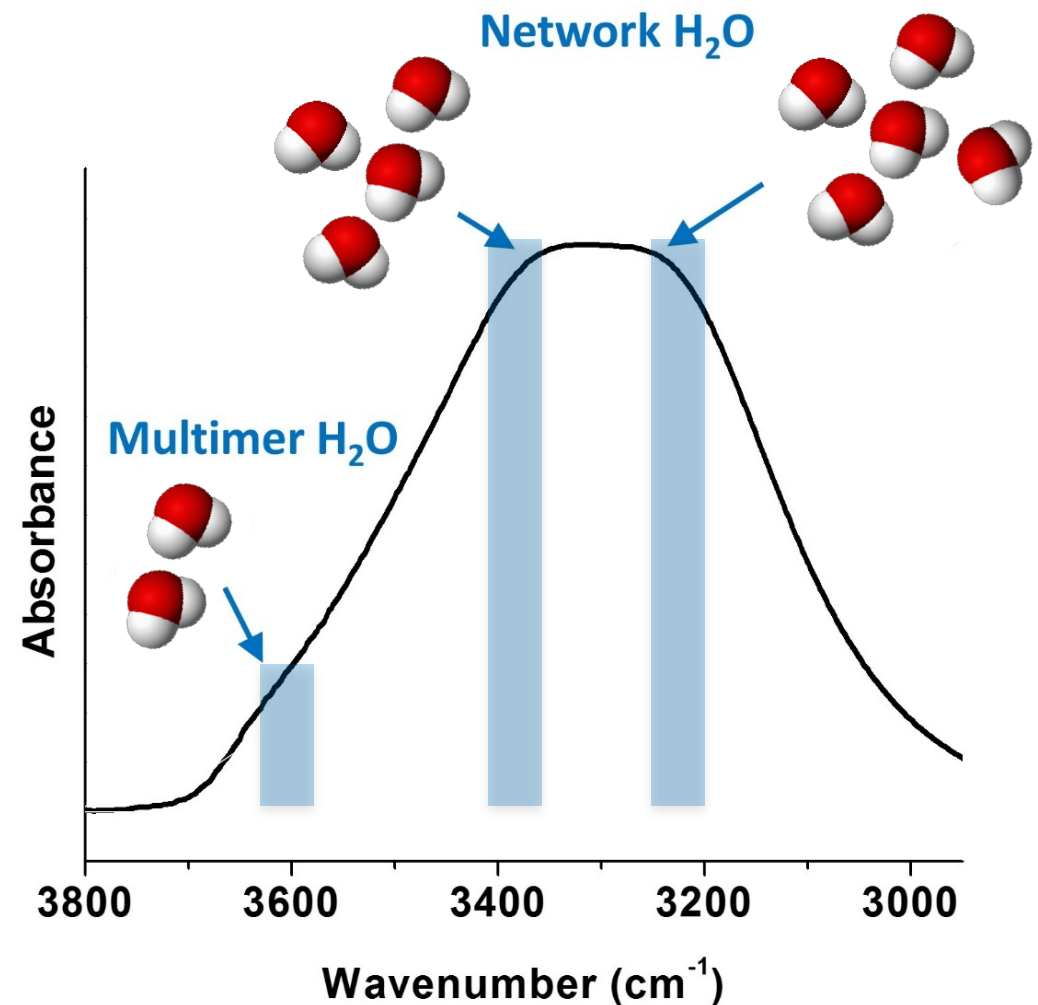
- deformation & colloidal glass [Shamana *et al.* (2017)]

IR Spectroscopy & Water Structure

- excellent technique to study water structure
 - O-H stretching mode of bulk water

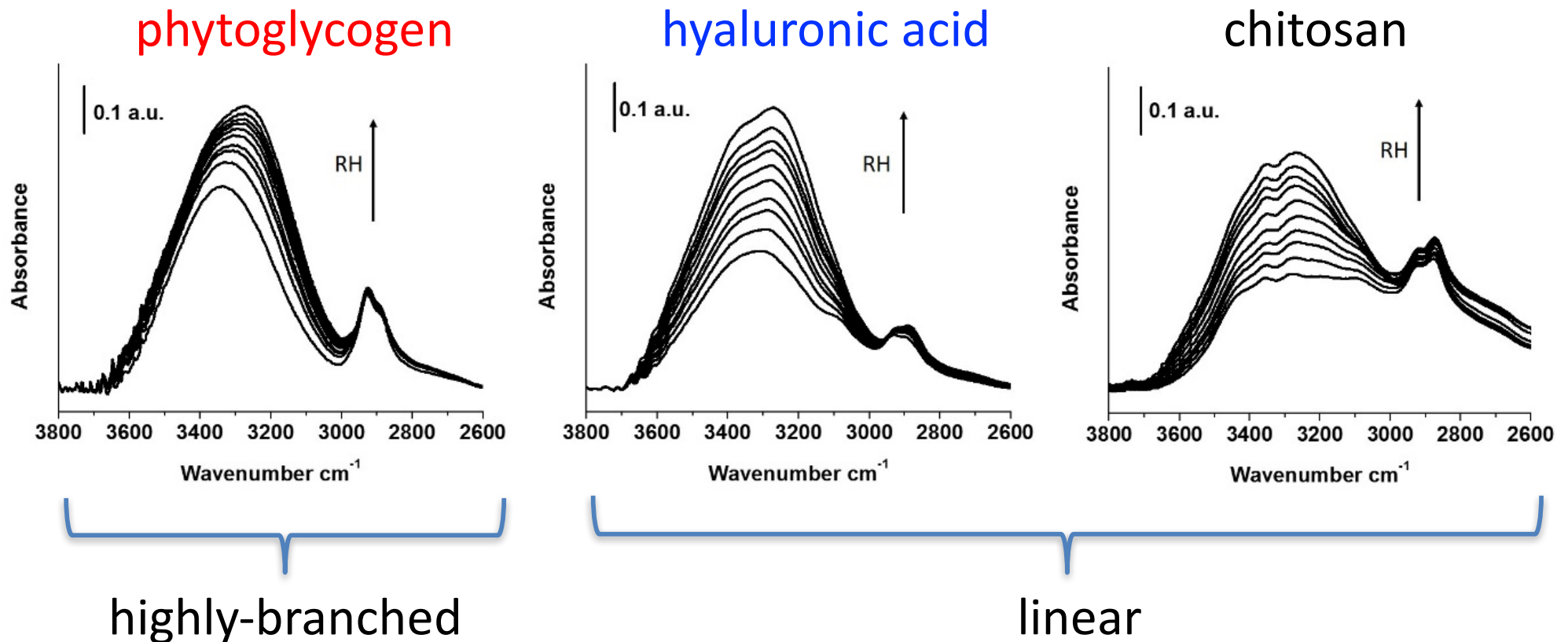


$$\nu_{\text{radiation}} = \nu_{\text{vibration}} = \frac{1}{2\pi} \sqrt{\frac{k_{\text{vibration}}}{m_{\text{vibration}}}}$$



Absorbance Spectra

- compare highly-branched phytyglycogen & two linear polysaccharides



[Grossutti & Dutcher, Biomacromolecules (2016)]

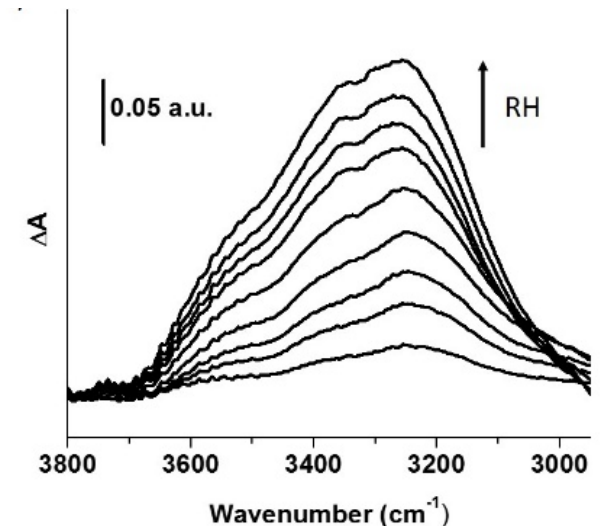
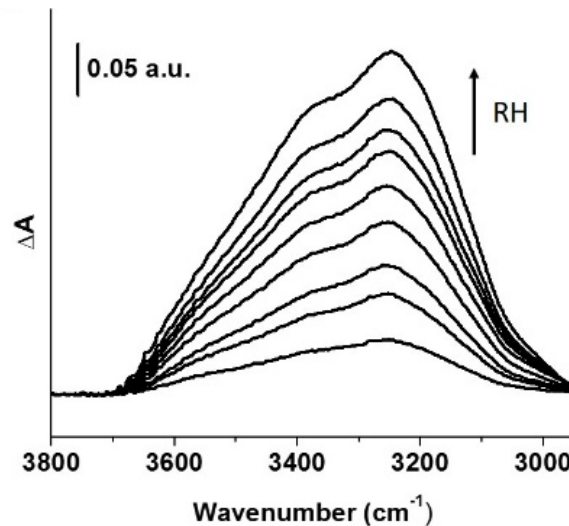
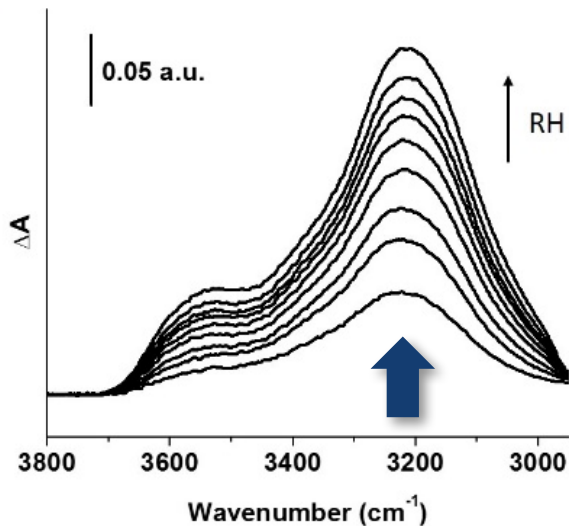
Difference Spectra

- difference between spectra at different RH & reference spectrum at lowest RH

phytoglycogen

hyaluronic acid

chitosan



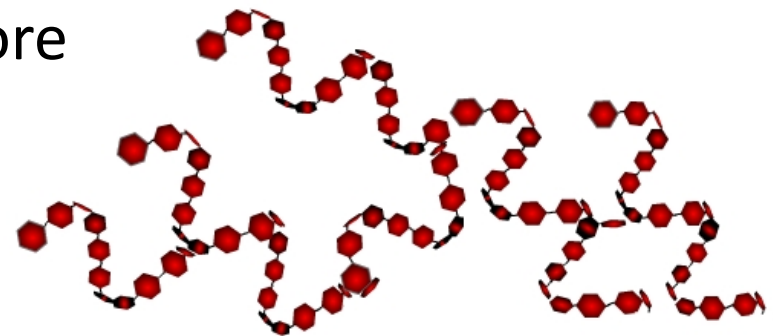
highly-branched

linear

- phytoglycogen has more highly connected, bound water
[Grossutti & Dutcher, Biomacromolecules (2016)]

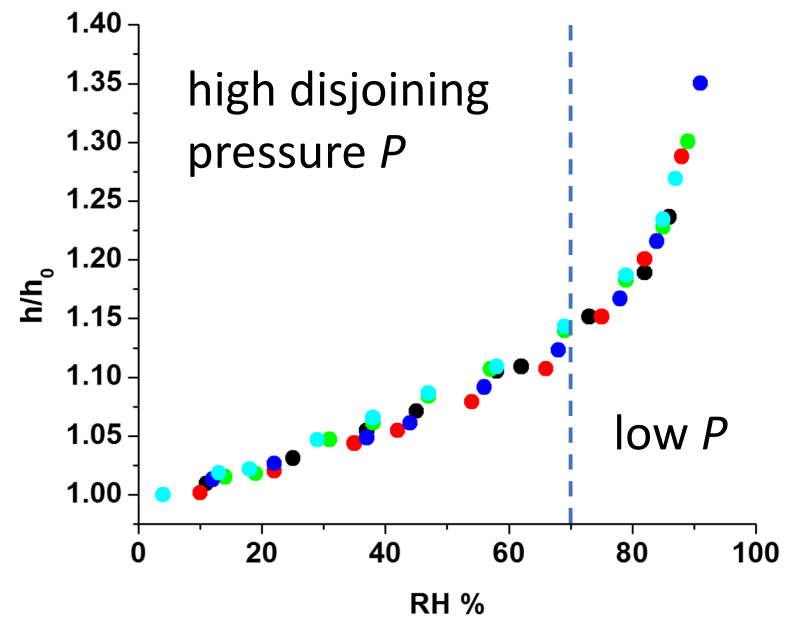
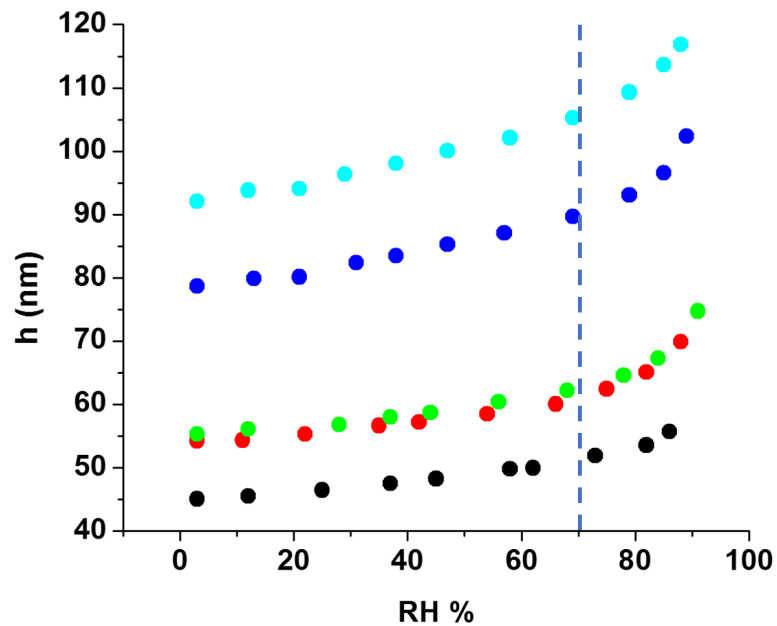
Nano-confined Water

- what is effect of nano-confinement on water structure?
- previous studies have shown
 - sensitivity to confinement
 - insensitivity to chemical environment
- generally introduces hydrogen bond defects & reduces water connectivity (polymers, micelles, liposomes)
- in contrast, connectivity increased for water confined in cylindrical glass nanopores
 - perhaps water in phytoglycogen more highly ordered because of water nanochannels along helical chains

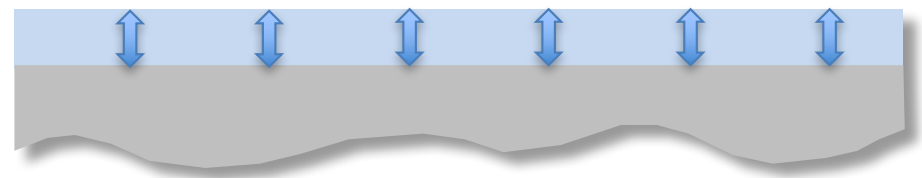


Ultrathin Film Swelling

- ellipsometry measurements of ultrathin phyto glycogen films



- interpret results in terms of disjoining pressure P acting across film

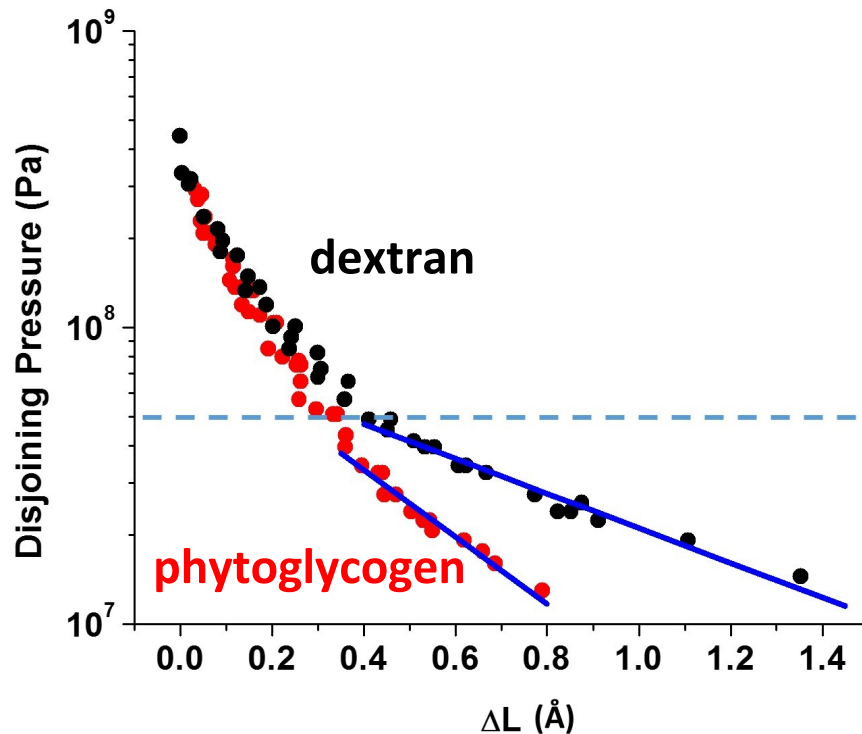


[Grossutti *et al.*, Langmuir (2017)]

Hydration Forces

- can relate swelling results at high RH to hydration forces

$$P(\Delta L) = P_0 e^{-\Delta L/\lambda_0} \quad \Delta L = a_0 \left[\left(\frac{h}{h_0} \right)^{1/3} - 1 \right]$$



compare glucose-based polysaccharides

polysaccharide	P_0 (10^7 Pa)	λ_0 (Å)
dextran	8.1 ± 0.5	1.71 ± 0.10
phytoglycogen	9.3 ± 0.8	0.89 ± 0.06
cellulose*	14	0.32



- decay length λ_0 depends on chain architecture

[Grossutti *et al.*, Langmuir (2017)]

*[Rehfeldt *et al.*, Langmuir (2003)]

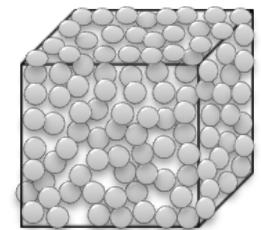
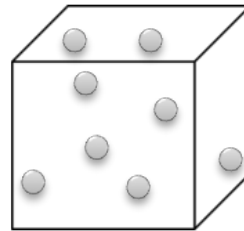
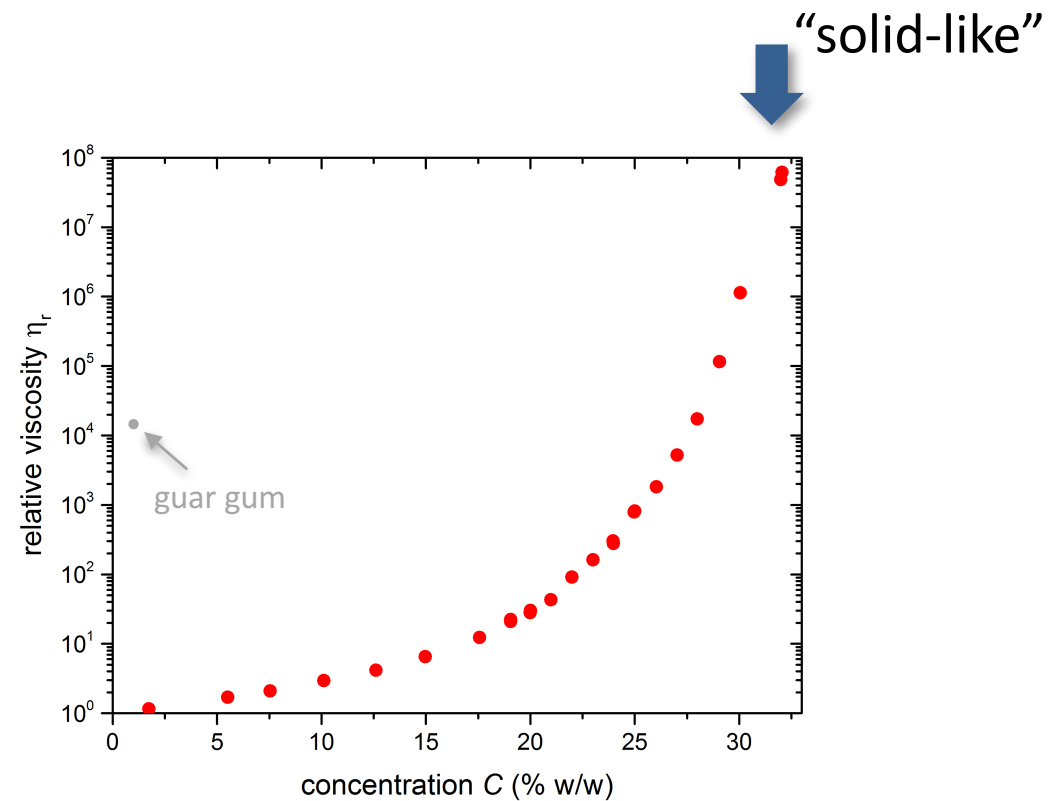
Phytoglycogen Particles In Water

- can be loaded up to 20% by mass before significant increase in viscosity



➔ “hard spheres”
(small interaction except when they touch)

[Shamana *et al.* (2017)]



Viscosity Measurements

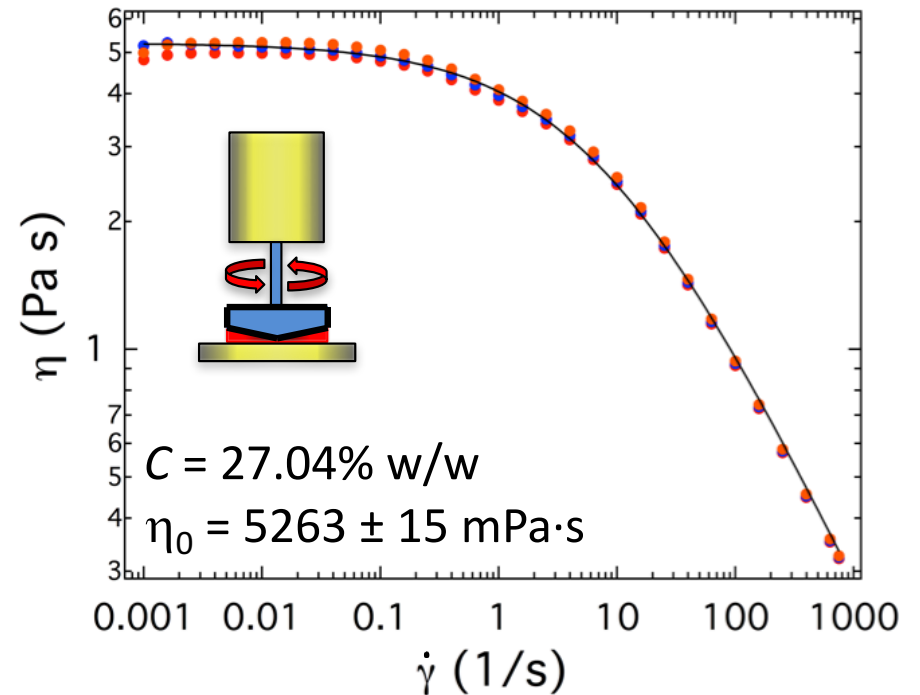
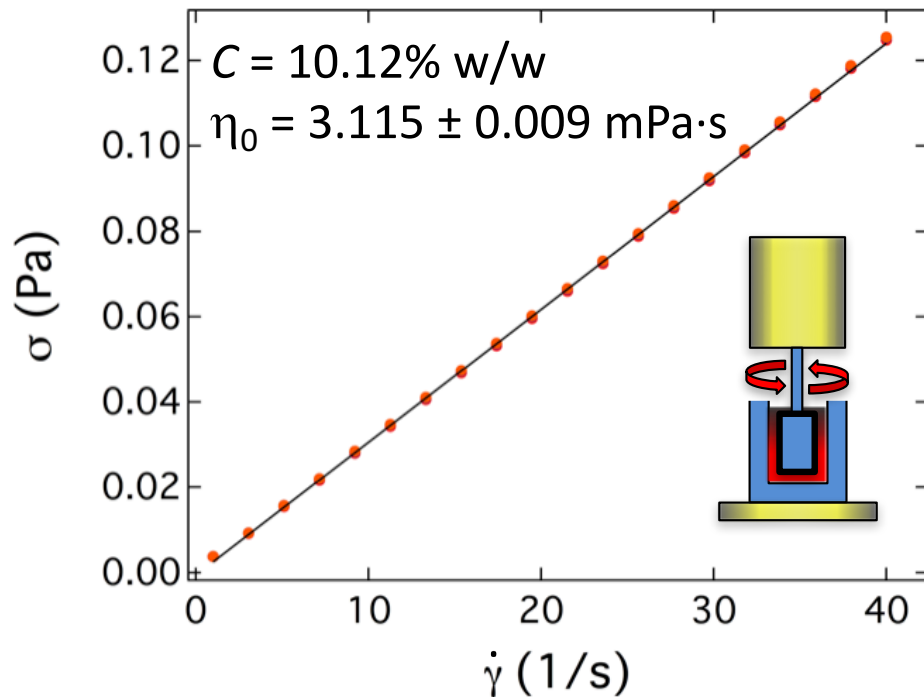
- mix spray dried particles at room temp into Milli-Q H₂O

$C \leq 20\%$ w/w: concentric cylinders

$C \geq 20\%$ w/w: cone & plate

$$\sigma = \eta \dot{\gamma} = \eta_0 \dot{\gamma}$$

$$\eta(\dot{\gamma}) = \frac{\eta_0}{1 + (k\dot{\gamma})^m}$$



Viscosity Measurements

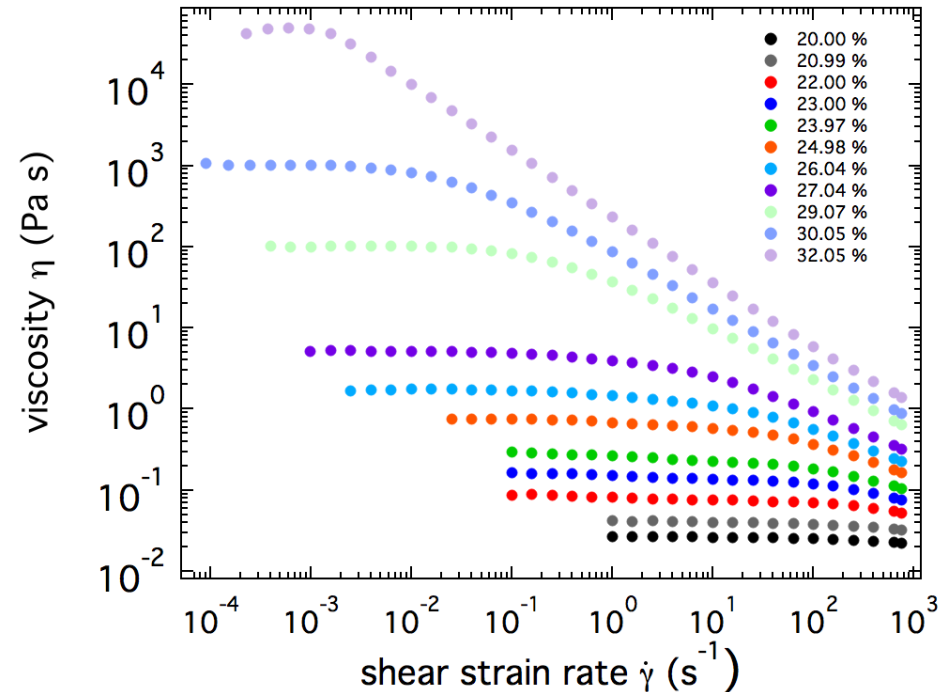
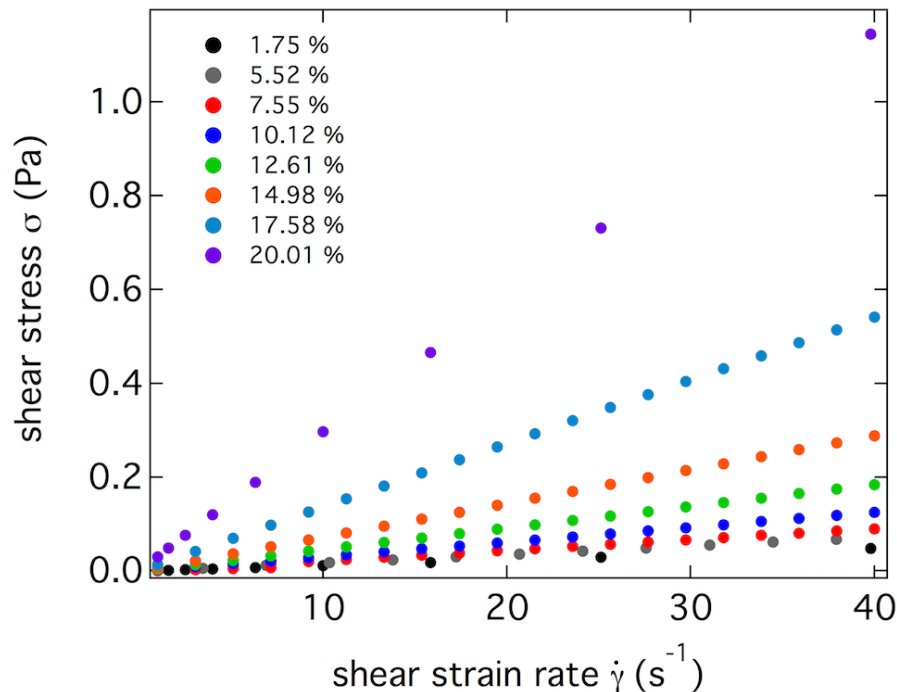
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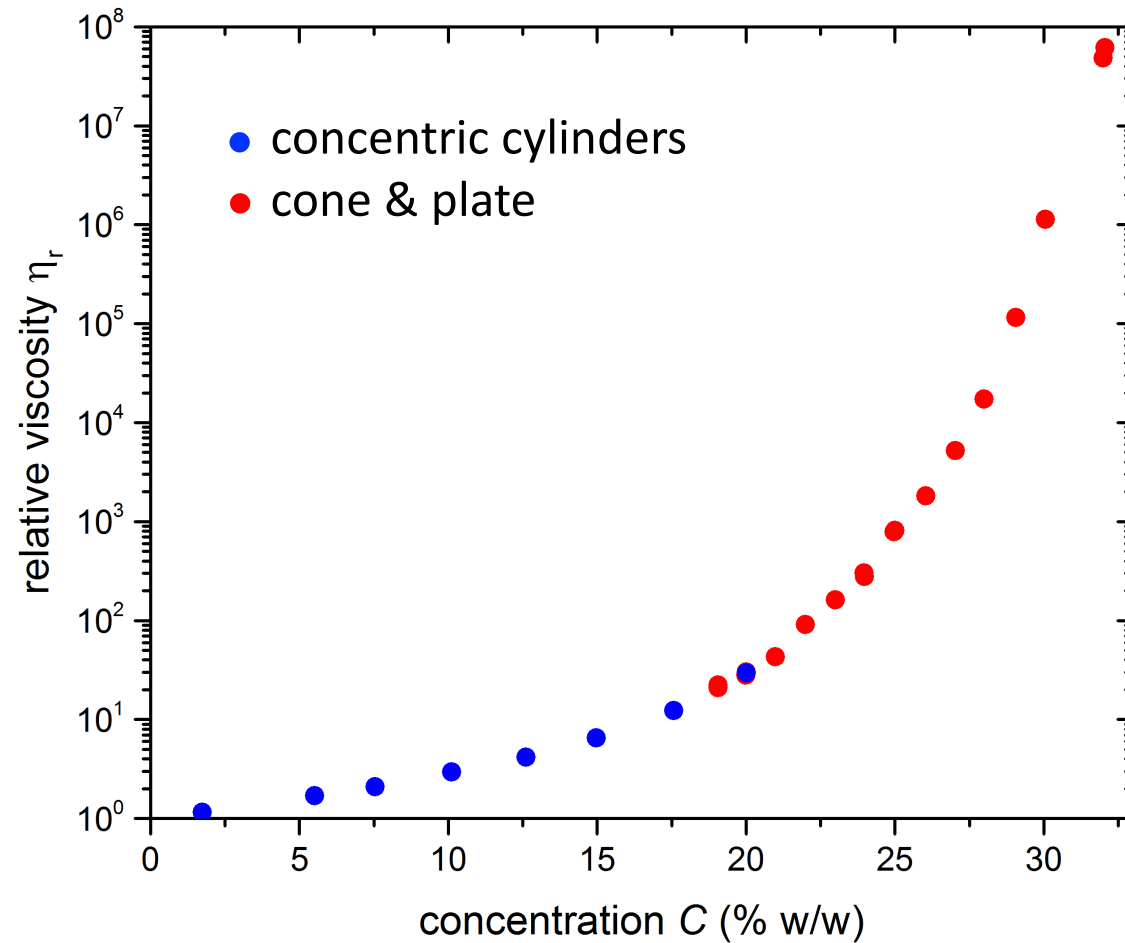
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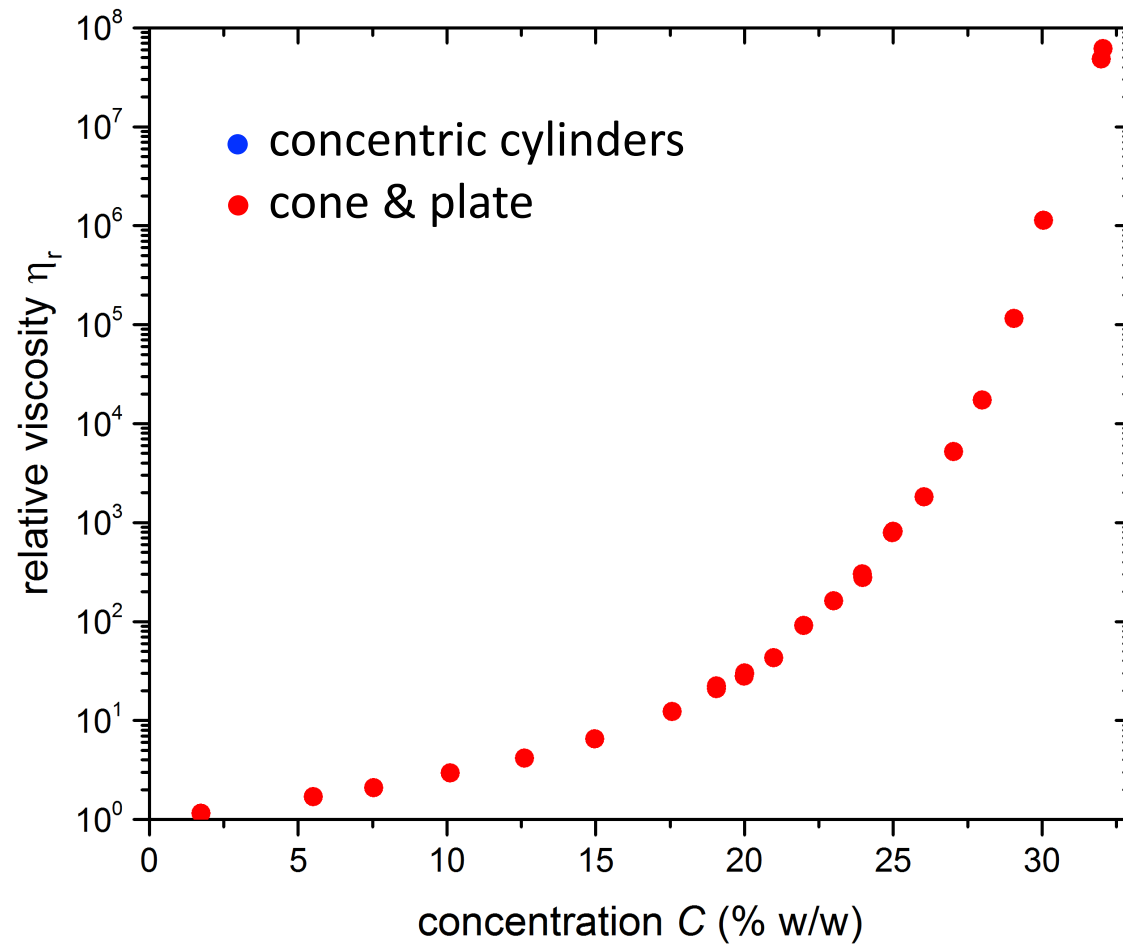
[Shamana *et al.* (2017)]

Zero Shear Viscosity



[Shamana *et al.* (2017)]

Zero Shear Viscosity



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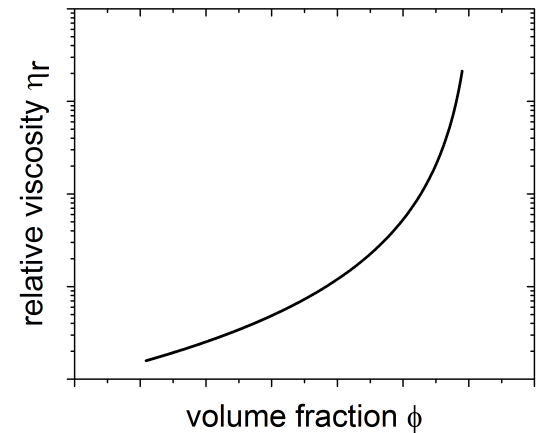
Hard Sphere Dispersion Theories

- properties of dispersion defined by volume fraction ϕ
- at low concentrations, viscosity described by Einstein relation

$$\eta_r = 1 + 2.5 \phi$$

- at high concentrations, semi-empirical approach [Krieger & Dougherty, Trans. Soc. Rheol. (1959)]

$$\eta_r = \left(1 - \frac{\phi}{\phi_m}\right)^{-[\eta]\phi_m}$$



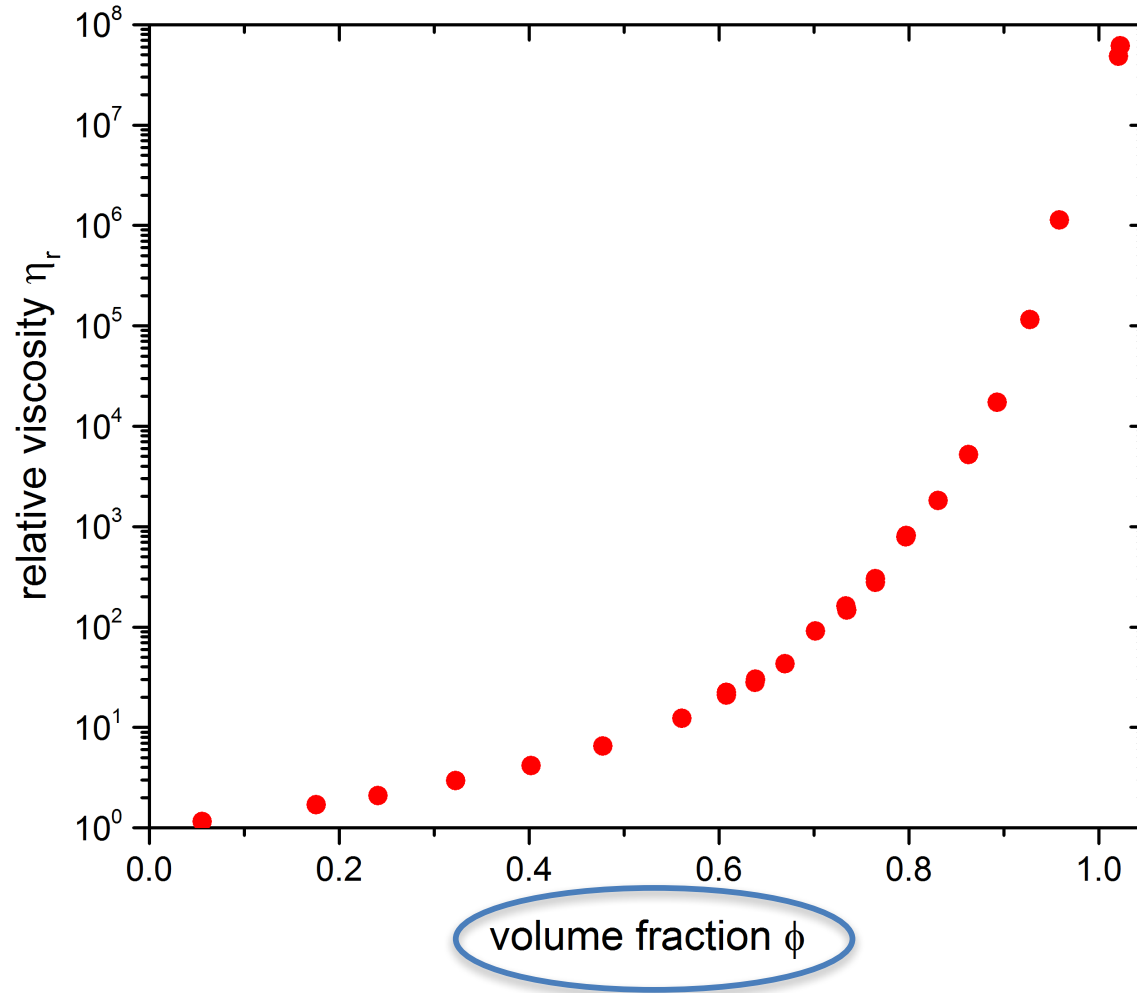
- describes divergence of viscosity due to packing of hard spheres at high volume fractions ϕ
- viscosity diverges at colloidal glass transition at $\phi_g = 0.58$ [Mason & Weitz, PRL (1995)]

Volume Fraction Values

- need to convert concentrations C into volume fractions ϕ
 - spray dried particles sorb considerable amount of water
- depends on quantities that we can measure
 - mass of spray dried particles
 - volume of added water
 - residual water in spray dried particles
 - density of hydrated particles

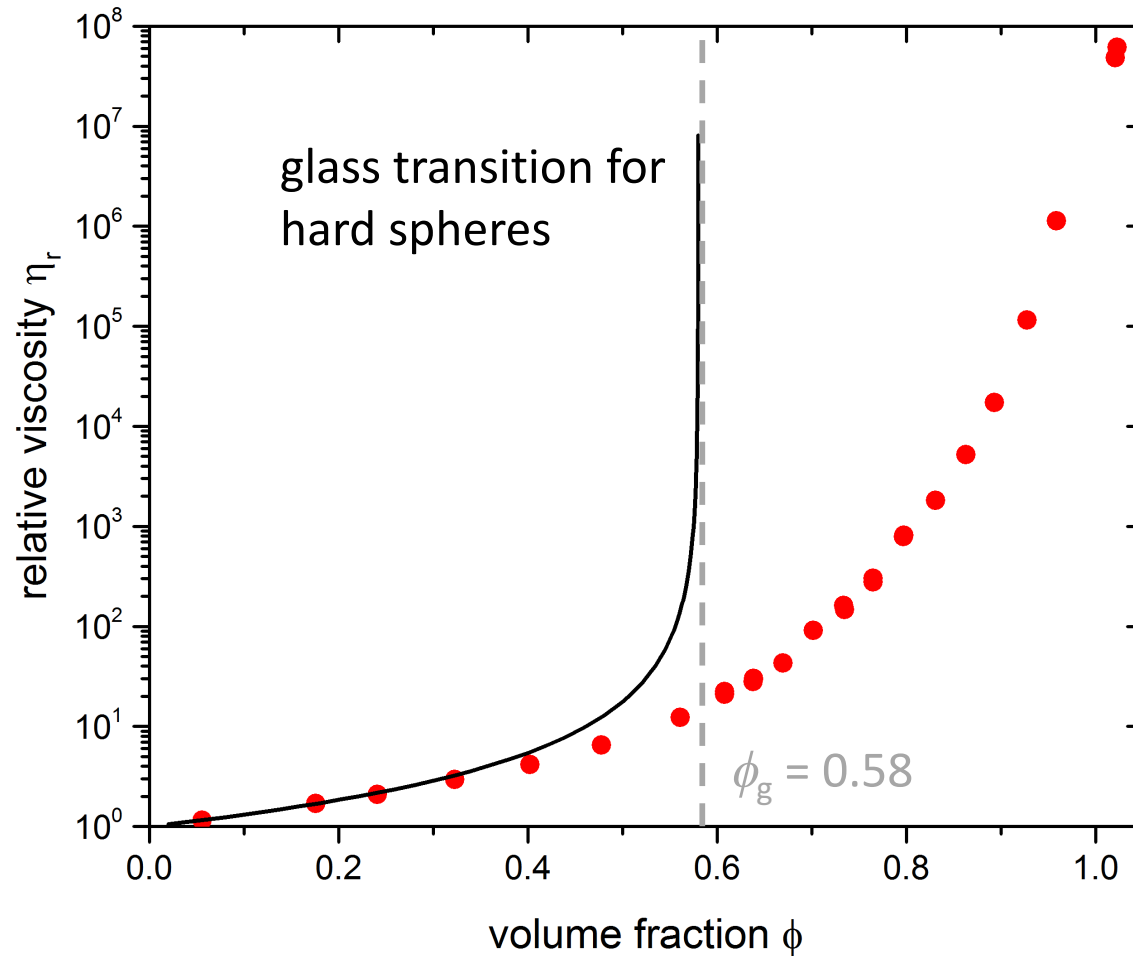
} measured
independently

Viscosity Versus Volume Fraction



[Shamana *et al.* (2017)]

Viscosity Versus Volume Fraction

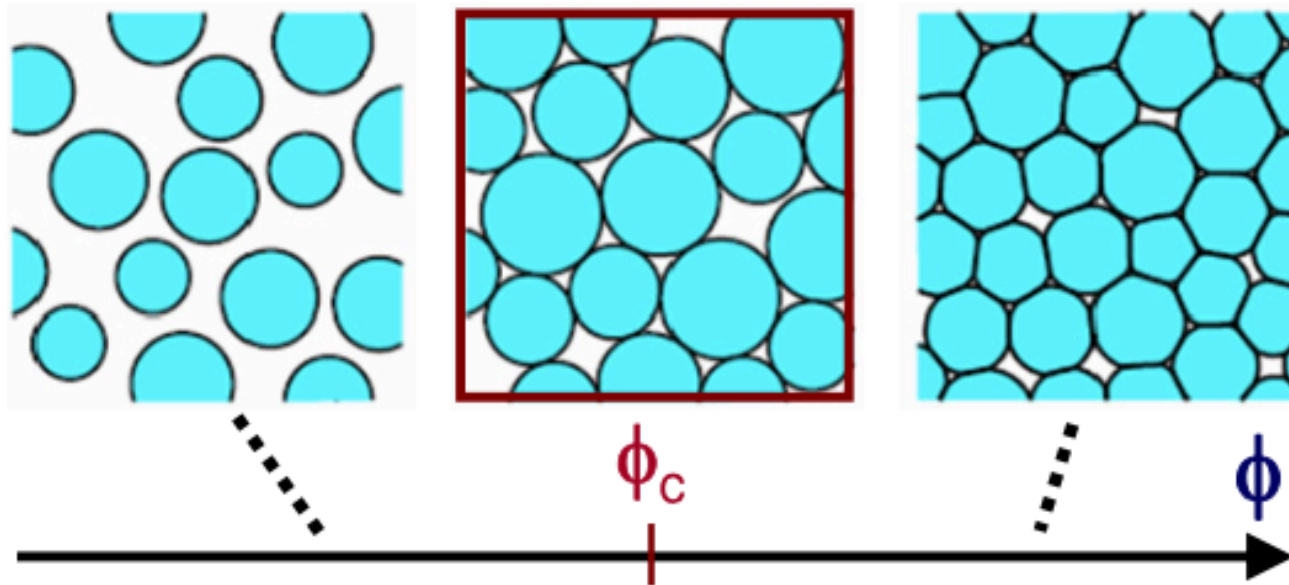


- considerable deformation of soft particles at high ϕ
 - consistent with neutron scattering & AFM [Nickels *et al.*, Biomacro (2016)]

[Shamana *et al.* (2017)]

Deformation at Large ϕ

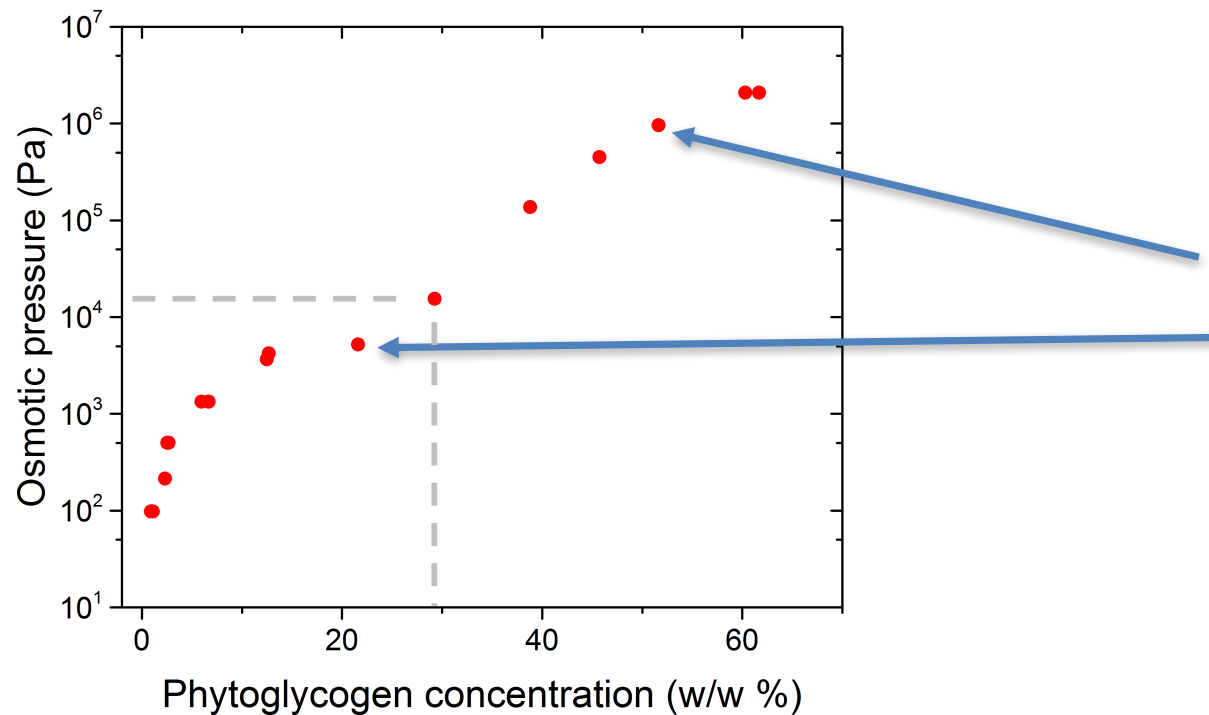
- soft particles will readily deform at large ϕ
 - facets will develop between particles to fill in voids up to $\phi = 1$ ($C \sim 30\%$ w/w)



[M. Van Hecke, J. Phys.: Condens. Matter (2010)]

Osmotic Pressure Study

- apply osmotic pressure Π by doing dialysis with PEG & dextran solutions & measure equilibrium concentration
 - achieve concentrations $> 60\%$ w/w

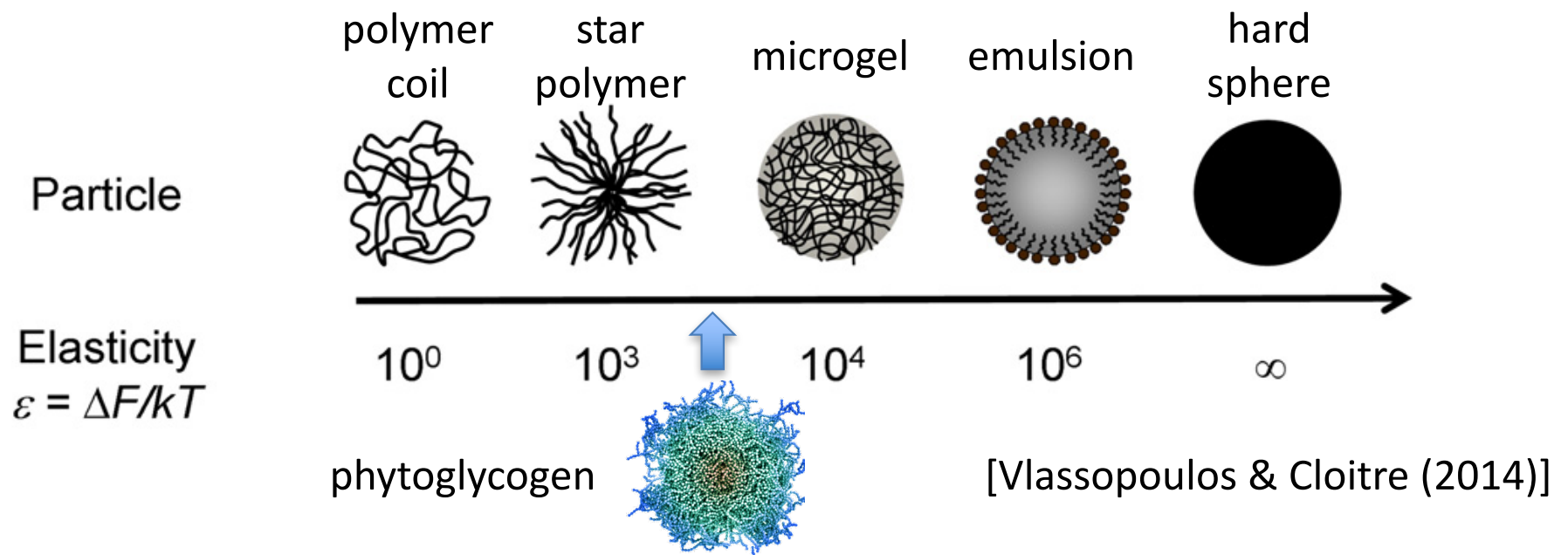


$$\Pi = 2 \times 10^6 \text{ Pa}$$

- for concentration $C = 30\%$ w/w, $\phi \sim 1$
 - value of Π provides estimate of bulk modulus of particles: $K \sim 20 \text{ kPa}$

Soft Colloid Systems

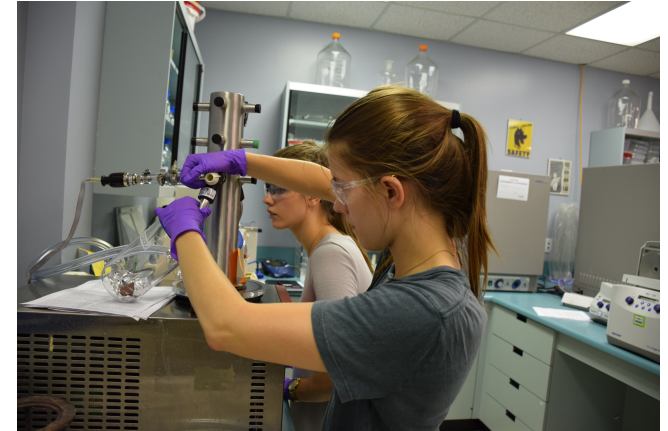
- soft colloid systems offer new insight into particle interactions & deformations



- phytoglycogen dispersions are a new model soft colloid system
 - native particles
 - chemically & enzymatically modified particles

PhytoSpherix™ Technology

- research in my U of Guelph laboratory
 - particles from non-GMO corn
 - successful crop trials
 - pilot scale production of tens of kg

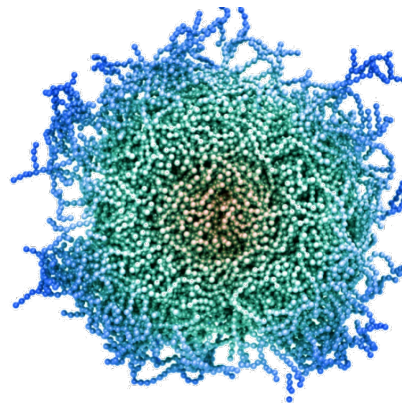


- IP protected by patent portfolio
- angel investment, \$7 M investment in 2015, \$12 M in 2017
- R&D facility in Guelph has 18 FT employees (9 PhD)
- new 12,000 ft² facility under construction, 16 T/y production
- approvals for use in cosmetics & food
- products developed, orders shipped, products available



Applications of PhytoSpherix Particles

- many advantages to using PhytoSpherix in personal care, nutritional & biomedical applications
 - water soluble
 - monodisperse
 - non-ionic
 - low viscosity
 - can be sterilized
 - easy to modify with high density of functional groups
- renewable resource
- biocompatible
- non-toxic
- non-allergenic
- biodegradable



Natural Multifunctional Ingredient



MIREXUS 



Personal Care

enhanced natural
cosmetics & anti-aging
products

Nutrition

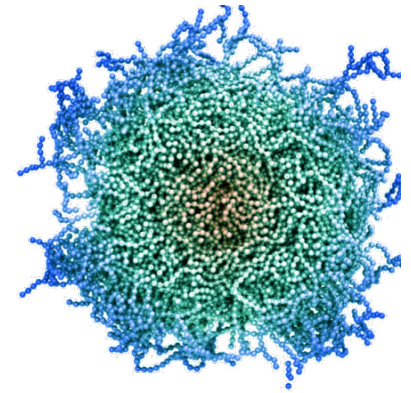
functional additives &
enhanced nutrition

Health Care

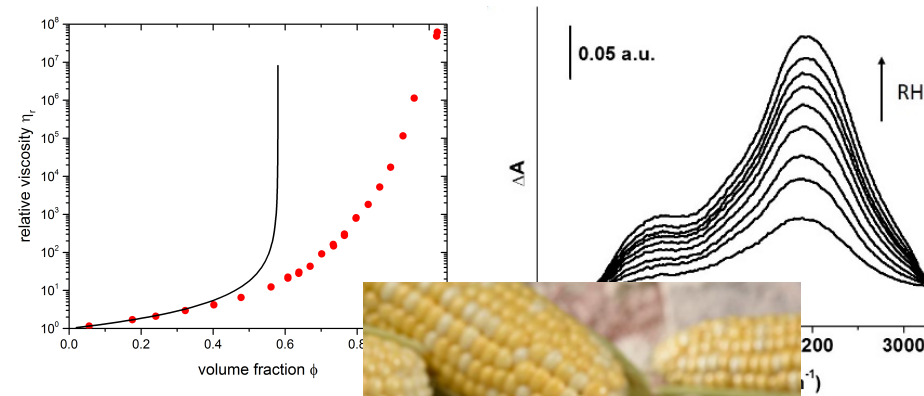
anti-infectives,
drug delivery &
immunomodulation

Phytoglycogen Nanoparticles

- nano from nature
 - safe, non-toxic – even edible!
 - special structure → special properties
 - unique strong interaction with water
 - model soft colloid system
- promising new material for fundamental research & new technologies



www.physics.uoguelph.ca/dutcherlab



The Team



www.mirexusbiotech.com



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