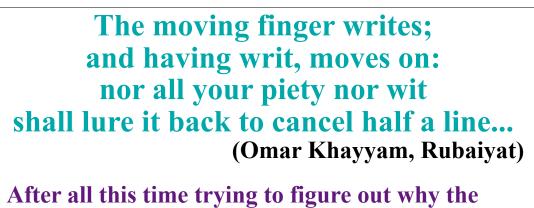
## FINAL EXAM SCHEDULED: Weds. Dec 12, 7pm - 10pm in BN2N (320 Huron, 2nd floor)

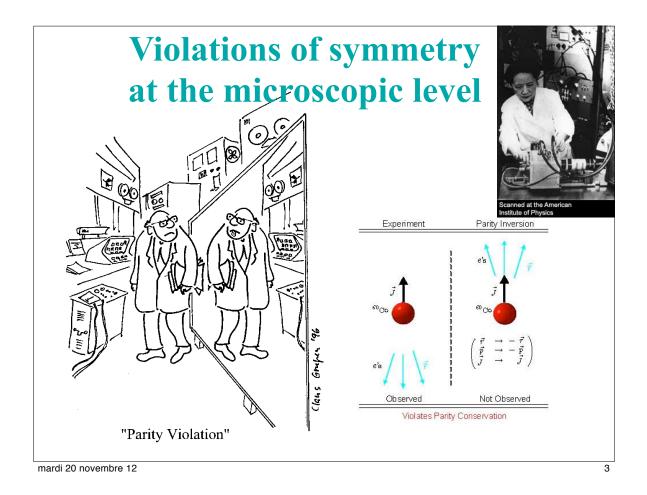
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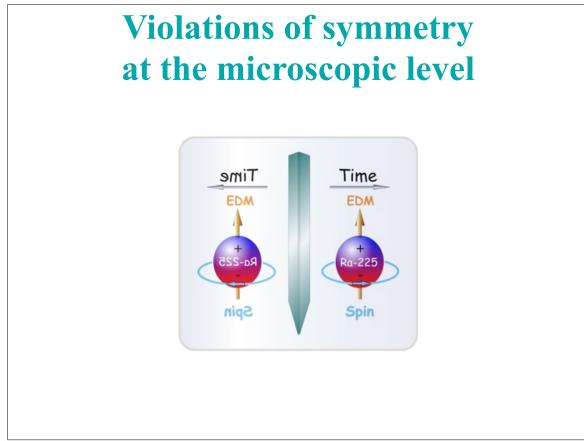


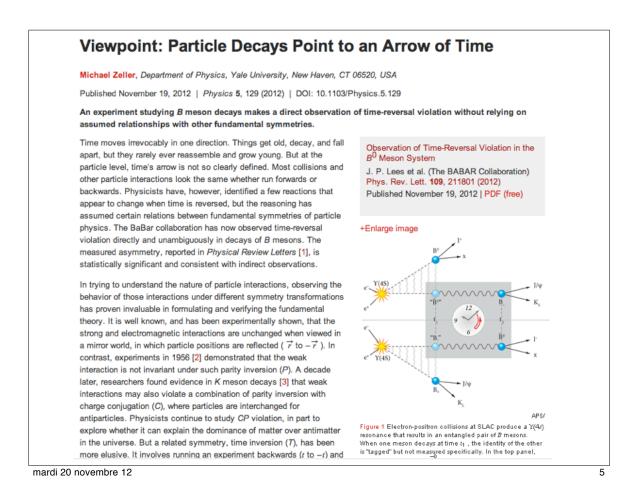
After all this time trying to figure out why the laws seemed symmetric when everyday life doesn't...

It turns out the laws *aren't* quite symmetric!

(and now we're even *more* surprised.)







 Several arrows of time (?)

 Entropy increases (water runs downhill, etc).

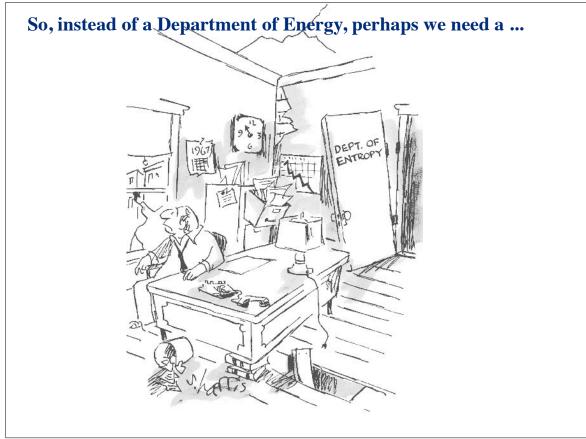
 We remember the past, not the future.

 The universe is expanding, not contracting.

 Schrödinger's cat is in a superposition before we observe him, but not after (?)

 Unclear to this day whether these are all related, or perhaps independent. Also unclear whether the microscopic violation of time-reversal symmetry is responsible for these macroscopic violations or not...

### Some earthly consequences • Mechanical energy can dissipate into heat, but the reverse doesn't occur. • Heat *can* be used to do work -- but always inefficiently. • A particular system may get more ordered with time - its entropy goes down. (When we cool water and it freezes into ice crystals, this happens.) Entropy puddle of water (no structure) minimum entropy maximum order maximum entropy But total entropy is going up -- to cool the water, we increase the overall entropy of the universe with our fridge's exhaust. • The "energy crisis" is really all about entropy. **Energy is conserved, but low-entropy, ordered systems** are hard to come by. mardi 20 novembre 12



# What is Quantum Mechanics?

The theory of atoms?

The theory of atoms and photons?

The theory of really small stuff?

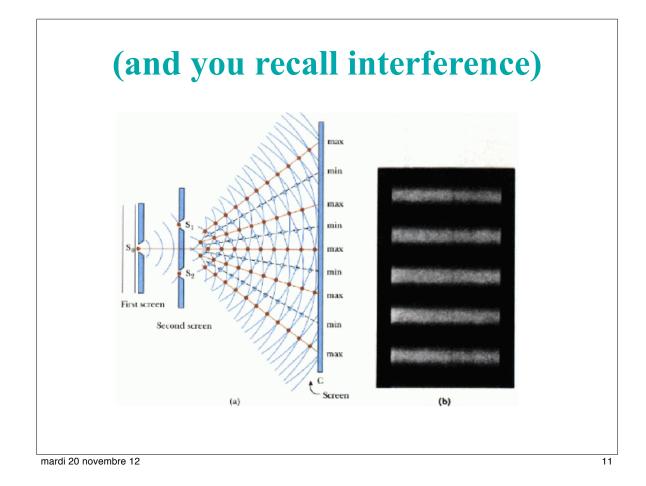
If not "the theory of everything," then at least the framework in which we (currently) believe the theory of everything would have to fit.

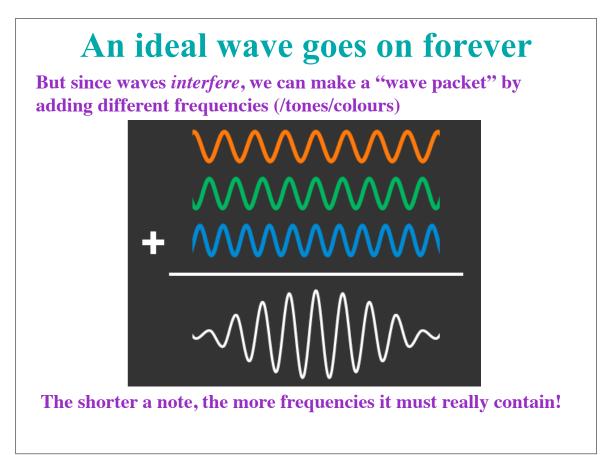
(I.e.: Einstein's relativity "corrected" Newton's laws -- but both discussed things that could be described as having a given position at a given time. QM says even the question is wrong -- reality is not *about* "positions" et cetera)

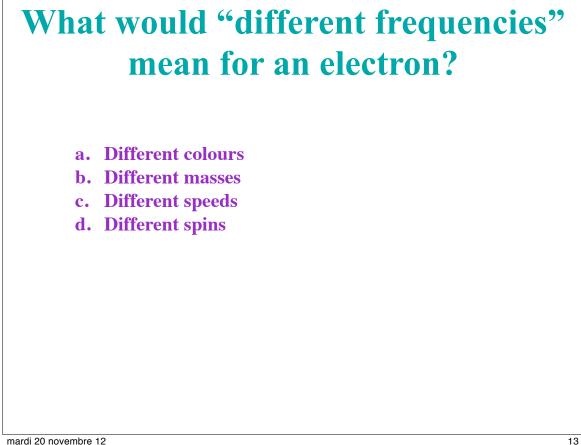
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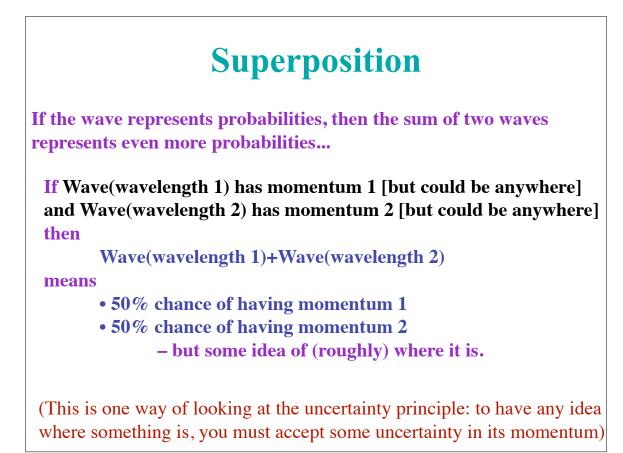
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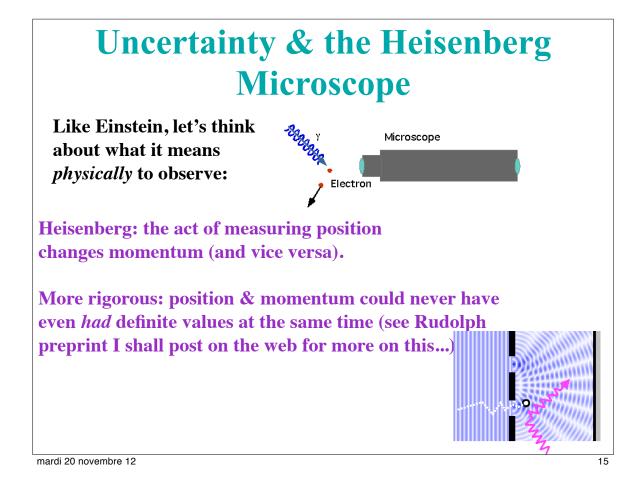


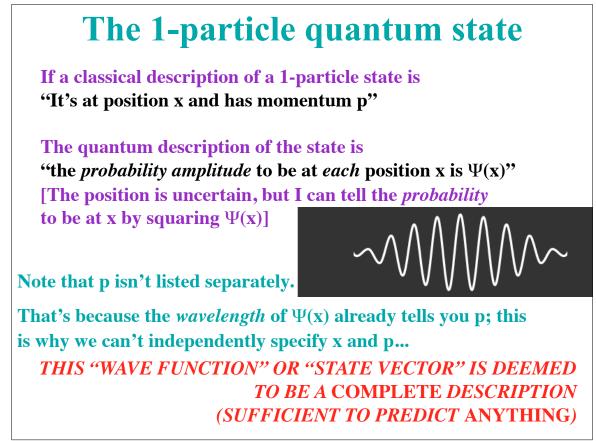


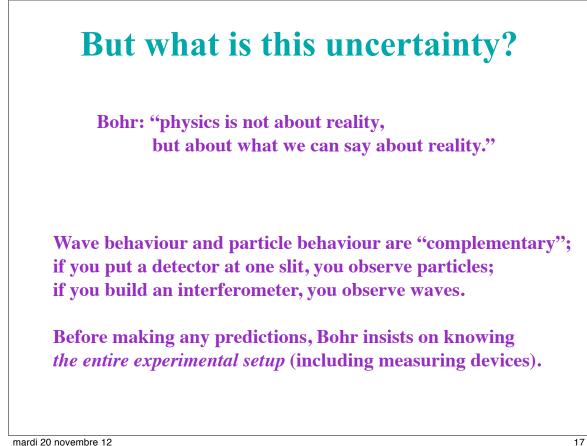


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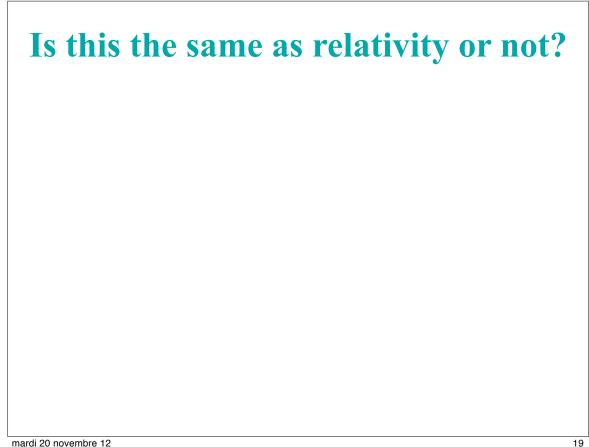
## **But what of reality??**

**Einstein:** Come on, there must be some meaning to position & momentum whether or not I measure them!



**Bohr:** Why, if you can't measure both? I'm just doing the same thing you did when you started relativity by pointing out "measurement is physical." You have to think about the measurement you're doing.

Einstein: a good joke shouldn't be repeated too often.



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## **Recall statistical thermodynamics**

Thermodynamics is an excellent theory, and predicts the expansion of heated gas, the efficiency of engines, et cetera; but it refuses to tell us *exactly where* any given air molecule is.

It is *correct*, but not *complete*.

Maybe QM is the same. A correct, *statistical*, description of how many identically prepared particles behave, but which is missing some "hidden variables" that tell us how each individual particle behaves...

## The general quantum state

If a classical description of a state of any system (the universe) is "If you measured all the measurable things -- position, momentum, and so on of every particle in existence -- they'd have the following list of values:  $p_1,x_1,p_2,x_2,...$ "

#### The quantum description of the state is

"the *probability amplitude* to be in each of the distinguishable states labelled by  $x_1, x_2, \dots$  is  $\Psi(x_1, x_2, \dots)$ "

Again:  $p_1, p_2, ...$  can't be listed separately. That's because the *wavelength* of  $\Psi(x_1, x_2, ...)$  already tells you the p's...

Note that there is no longer a separate description of "the reality for particle 1" and "the reality for particle 2" -- there is one big function for the probability of *everything*. We will keep returning to this issue of "entanglement," the oddest thing about QM...

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# So what happens when I measure something?

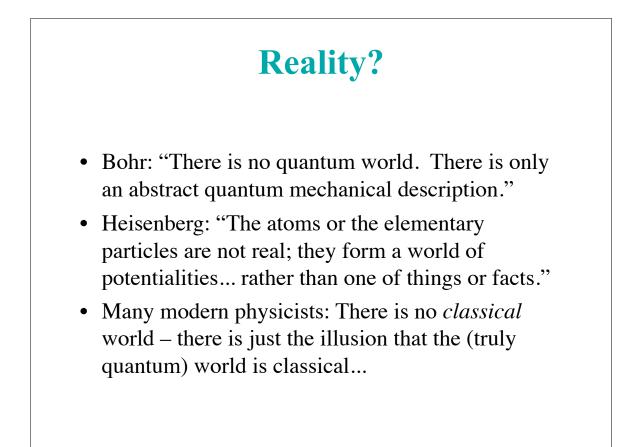
The position and velocity couldn't have both been known, but I get an answer whichever one I measure... and even if I measure position, where will it be a minute later??

The problem we're still arguing about today: What does it mean that it's "impossible" to know both?

- Particles really have definite positions & momenta, but we don't know how to measure them? And QM is just a theory of the "big picture," like thermodynamics?
- Particles don't actually have definite positions & momenta? (or any other definite properties, for that matter?)

The quantum state "collapses" randomly when we look at it?





# A few possible perspectives...

## Copenhagen interpretation

From Wikipedia, the free encyclopedia

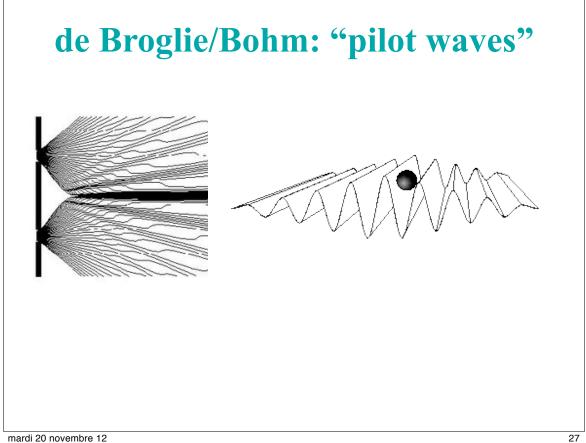
The **Copenhagen interpretation** is one of the earliest and most commonly taught interpretations of quantum mechanics.<sup>[1]</sup> It holds that quantum mechanics does not yield a description of an objective reality but deals only with probabilities of observing, or measuring, various aspects of energy quanta, entities which fit neither the classical idea of particles nor the classical idea of waves. According to the interpretation, the act of measurement causes the set of probabilities to immediately and randomly assume only one of the possible values. This feature of the mathematics is known as wavefunction collapse. The essential concepts of the interpretation were devised by Niels Bohr, Werner Heisenberg and others in the years 1924–27.

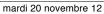
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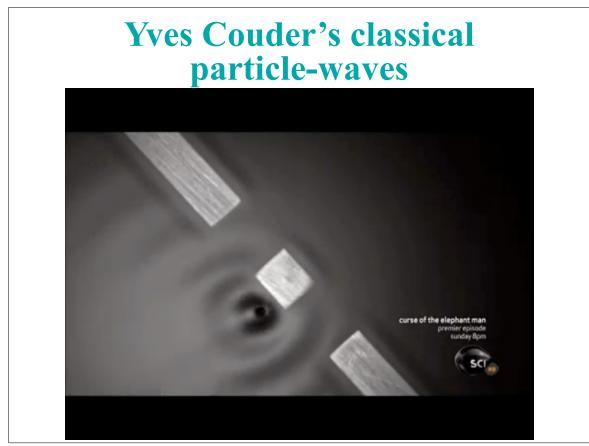
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## Wiki's view of Copenhagen...

- 1. A system is completely described by a wave function  $\psi$ , representing the state of the system.
- 2. The description of nature is essentially probabilistic, with the probability of an event related to the square of the amplitude of the wave function related to it. (The Born rule, after Max Born)
- It is not possible to know the value of all the properties of the system at the same time; those properties that are not known with precision must be described by probabilities. (Heisenberg's uncertainty principle)
- 4. Matter exhibits a wave-particle duality. An experiment can show the particle-like properties of matter, or the wave-like properties; in some experiments both of these complementary viewpoints must be invoked to explain the results, according to the complementarity principle of Niels Bohr.
- Measuring devices are essentially classical devices, and measure only classical properties such as position and momentum.
- 6. The quantum mechanical description of large systems will closely approximate the classical description. (The correspondence principle of Bohr and Heisenberg.)







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