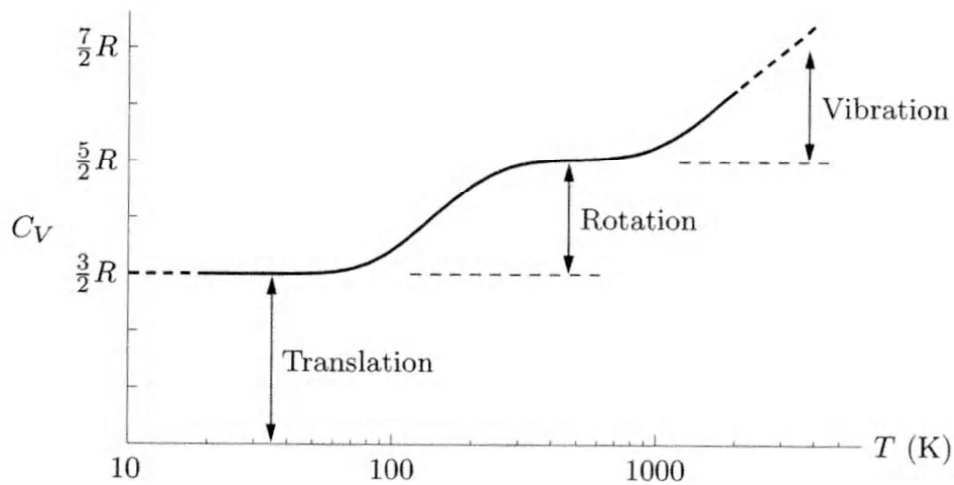


Frozen degrees of freedom

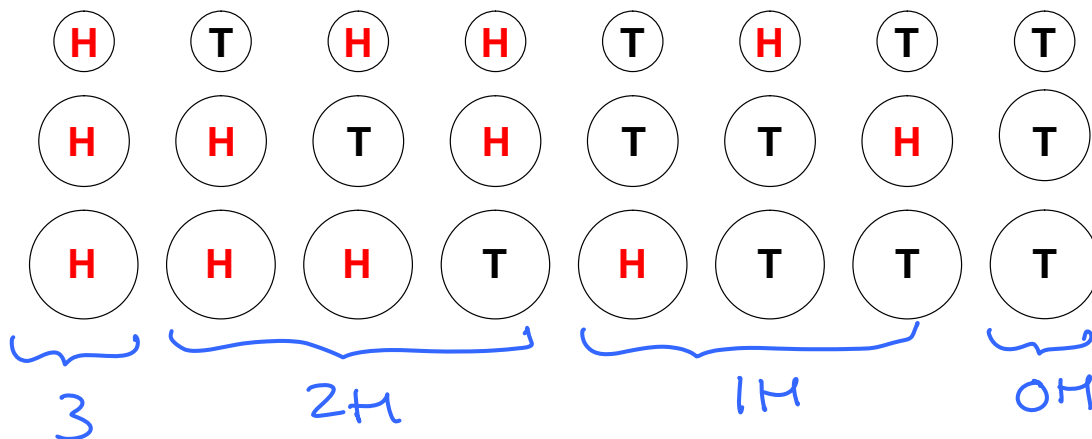
- Example of H_2 molecule



- Need Quantum and Statistical Mechanics!

Example - Coin flip

- Starting with a dime, a quarter and a loonie:
 - Flip all three - what are possible outcomes?



Some definitions for these “States”

microstate \rightarrow specify state of each particle

macrostate \rightarrow overall state (e.g. HHT)

multiplicity \rightarrow number of microstates that correspond to a given macrostate

$$\Omega(3H) = 1 \quad \text{or} \quad \Omega(2H) = 3$$

$$\text{total multiplicity} \quad \Omega(\text{all}) = 8$$

probability of getting particular macrostate

$$(n \text{ heads}) = \frac{\text{multiplicity}}{\text{total multiplicity}} = \frac{\Omega(n)}{\Omega(\text{all})}$$

Considering a larger system...

- If you have 10 coins, how do you calculate $\Omega(n)$?

– 0H: $\Omega(0) = 1$

1 2 3 4 5 6 7 8 9 10

– 1H: $\Omega(1) = 10$

1 2 3 4 5 6 7 8 9 10

– 2H: $\Omega(2) = \frac{10 \times 9}{2}$

1 2 3 4 5 6 7 8 9 10

– 3H: $\Omega(3) = \frac{10 \times 9 \times 8}{3 \times 2 \times 1}$

1 2 3 4 5 6 7 8 9 10

1 2 3 4 5 6 7 8 9 10

1 2 3 4 5 6 7 8 9 10

1 2 3 4 5 6 7 8 9 10

1 2 3 4 5 6 7 8 9 10

1 2 3 4 5 6 7 8 9 10

$$= \frac{10 \times 9 \times 8 \times 7 \times \dots \times 1}{(7 \times 6 \times 5 \times \dots \times 1)(3 \times 2 \times 1)}$$

$$= \frac{10!}{7! 3!}$$

Multiplicity of a Macrostate

- Generalizing from 10 to N

$$\Omega(3) = \frac{10!}{7!3!} \Rightarrow \Omega(n) = \frac{N!}{n!(N-n)!} = \binom{N}{n}$$

N total number of coins

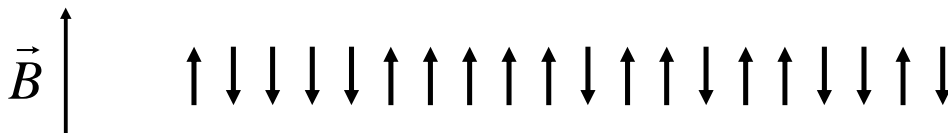
n number of coins chosen

$\Omega(n) \Rightarrow$ number of ways of choosing n objects out of N

Moving to a Model System

Another example is the two-state paramagnet

- In a paramagnetic material
 - Magnetic “dipoles” tend to align parallel to any externally applied magnetic field
- Quantized - so only certain discrete values are allowed
 - Simplest case, only two values are allowed
 - Either positive or negative orientation of dipole



Two-state Paramagnet

- At a particular time,

N_{\uparrow} upward N_{\downarrow} downward

$$N = N_{\uparrow} + N_{\downarrow}$$

one macrostate for each N_{\uparrow} from $0 \rightarrow N$

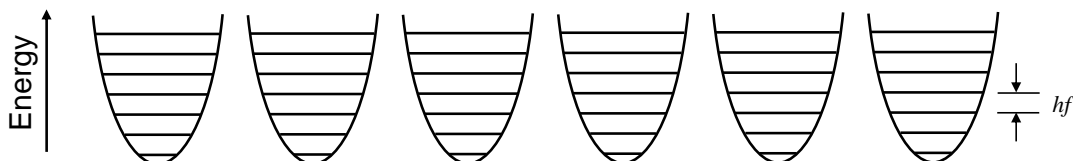
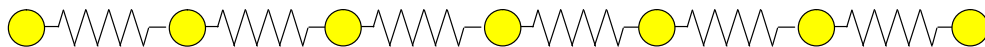
multiplicity of macrostate

$$\Omega(N_{\uparrow}) = \binom{N}{N_{\uparrow}} = \frac{N!}{N_{\uparrow}! N_{\downarrow}!}$$

A more complex system

Moving to a more realistic or “physical” model system

- A set of microscopic systems that can store any number of energy “units”, all of the same size
 - Such as quantum-mechanical harmonic oscillator
 - Potential energy function: $\frac{1}{2} k_s x^2$
 - Size of all energy “units”: hf

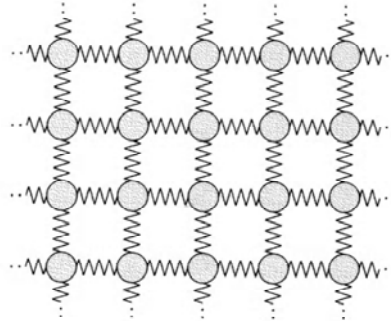


$h = \text{Planck's constant}$
 $= 6.63 \times 10^{-34} \text{ J}\cdot\text{s}$

$f = \text{natural frequency of oscillator}$
 $= \frac{1}{2\pi} \sqrt{\frac{k_s}{m}}$

Einstein Model of a Solid

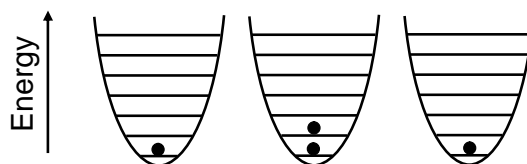
- Oscillations of atoms in a solid can be described as a collection of identical oscillators
 - In three-dimensional solid, can vibrate in three independent directions
 - N oscillations \Rightarrow only $N/3$ atoms



Einstein Solid

- Each oscillator can have multiple energy units
 - Example: 3 oscillators

- 0 units of energy ($0hf$): $\Omega(0) = 1$
- 1 unit of energy (hf): $\Omega(1) = 3$
- 2 units of energy ($2hf$): $\Omega(2) = 6$ $(2,0,0)$ $(1,1,0)$
- 3 units of energy ($3hf$): $\Omega(3) = 10$
 $(3,0,0)$ $(2,1,0)$ $(1,1,1)$



Larger Einstein Solid

- In general, for N oscillators with q energy units:

$$\Omega(N, q) = \binom{q+N-1}{q} = \frac{(q+N-1)!}{q!(N-1)!}$$

using • energy unit
| line for partition $N-1$

for oscillators

• | • | • (1, 1, 1,)

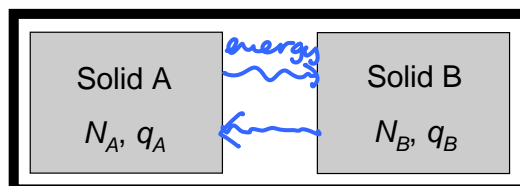
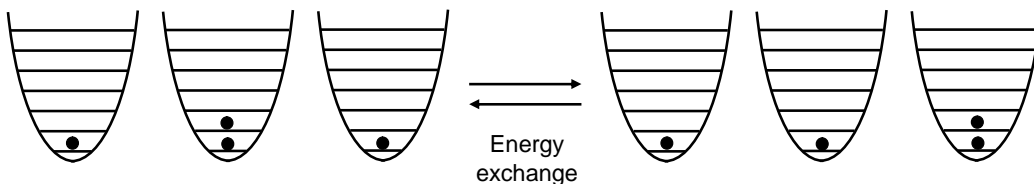
• • | | • (2, 0, 1)

↑ 5 other arrangements of this microstate

Interacting Systems

Two Einstein solids that can share energy back and forth - A & B

- Assuming weakly coupled systems
 - Exchange of energy between atoms in each solid is much faster than between the two - U_A and U_B



System of Two Einstein Solids

- Each solid has 3 harmonic oscillators (N_A, N_B) and they contain a total of six units of energy (q_{total})

$$N_A = N_B = 3$$

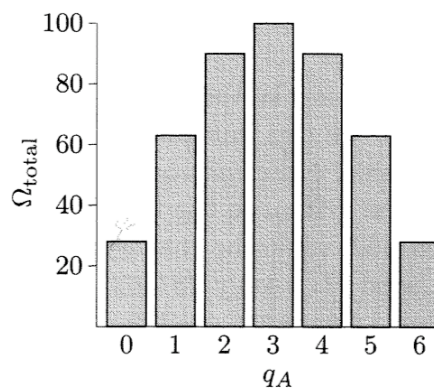
$$q_{total} = q_A + q_B = 6$$

to calculate energy

need $U = q h f$ ← harmonic oscillator

System of Two Einstein Solids

q_A	Ω_A	q_B	Ω_B	$\Omega_{total} = \Omega_A \Omega_B$
0	1	6	28	28
1	3	5	21	63
2	6	4	15	90
3	10	3	10	100
4	15	2	6	90
5	21	1	3	63
6	28	0	1	28
				$462 = \binom{6+6-1}{6}$

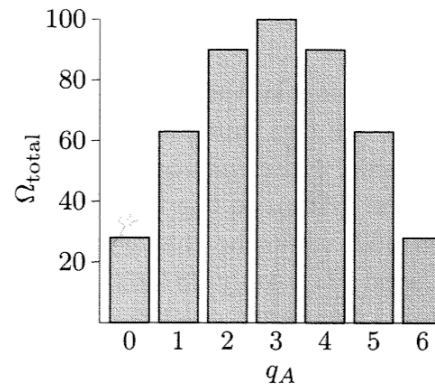


Ω_{total} depend on number of ways of choosing A & B $\Omega = \binom{q+N-1}{q}$

for each of Ω_A microstates of A there are Ω_B microstates available for B

System of Two Einstein Solids

q_A	Ω_A	q_B	Ω_B	$\Omega_{\text{total}} = \Omega_A \Omega_B$
0	1	6	28	28
1	3	5	21	63
2	6	4	15	90
3	10	3	10	100
4	15	2	6	90
5	21	1	3	63
6	28	0	1	28
				$462 = \binom{6+6-1}{6}$



- Fundamental assumption of Statistical Mechanics
 - In an isolated system in thermal equilibrium, all accessible microstates are equally probable

Probabilities of Macrostates

- If all *microstates* are equally probable, some *macrostates* will be more probable than others