

Review

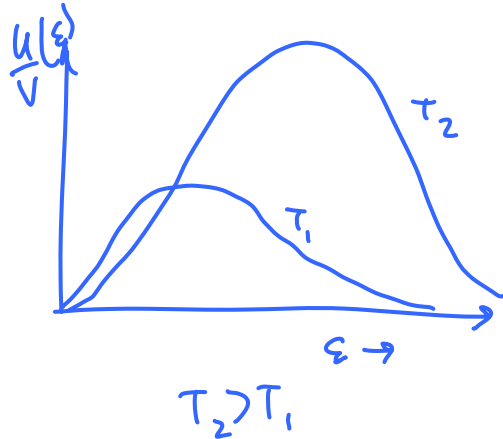
Blackbody radiation - quantum statistical description needed

- Planck distribution $\bar{n}_{Pl} = \frac{1}{e^{hf/kT} - 1}$

← number of photons of given frequency

- Planck spectrum $\frac{8\pi}{(hc)^3} \frac{\epsilon^3}{e^{\epsilon/kT} - 1}$

energy density per photon energy $\rightarrow \frac{U}{V}(\epsilon)$
 photon energy $\epsilon = hf$



Heat Capacity and Entropy

- Total energy density found by integrating over all energy units

$$\frac{U}{V} = \frac{8\pi}{(hc)^3} \int_0^{\infty} \frac{\epsilon^3}{e^{\epsilon/kT} - 1} d\epsilon \quad x \equiv \frac{\epsilon}{kT}$$

$$\frac{U}{V} = \frac{8\pi}{(hc)^3} (kT)^4 \int_0^{\infty} \frac{x^3}{e^x - 1} dx \quad \rightarrow \frac{\pi^4}{15}$$

$$U = \underbrace{\frac{8\pi^5 k^4 V}{15 \cdot h^3 \cdot c^3}}_a T^4 = a T^4$$

heat capacity

$$C_v = \left(\frac{\partial U}{\partial T} \right)_v = 4aT^3 \propto VT^3$$

absolute entropy

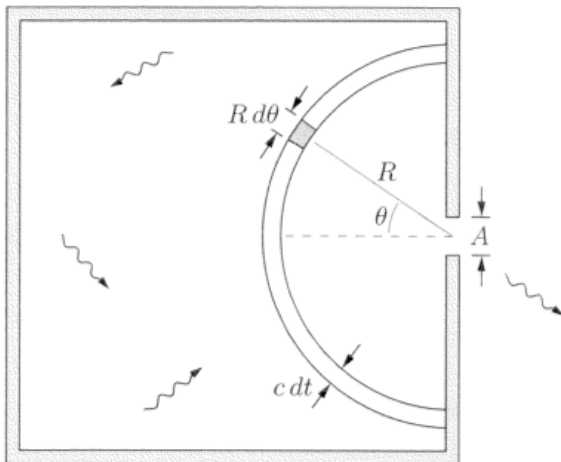
$$S(T) = \int_0^T \frac{C_v(T')}{T'} dT' = \frac{4}{3}aT^3$$

Stefan's Law

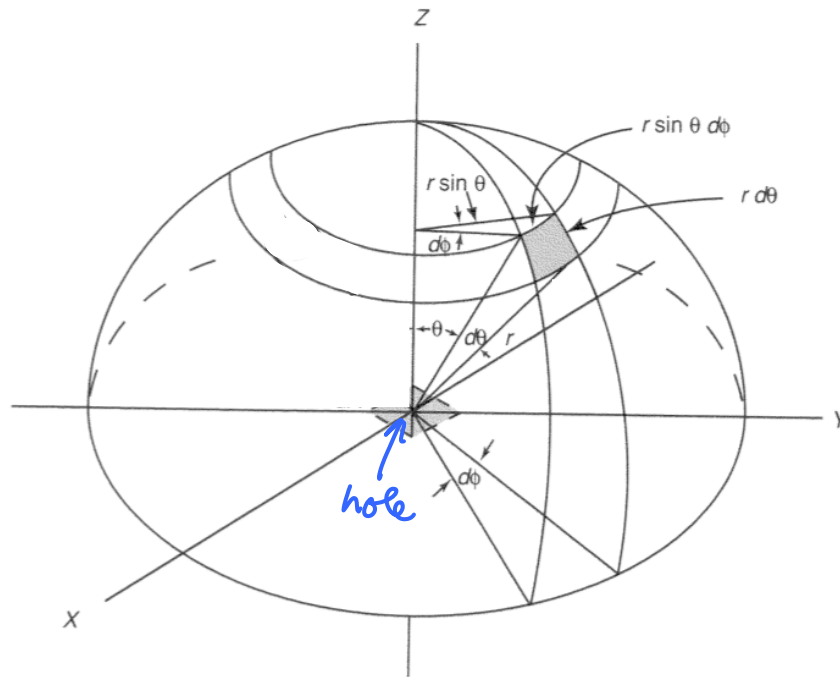
- Relates radiated power to temperature

↳ rate of energy transfer

- Consider photons escaping from the box from small hole



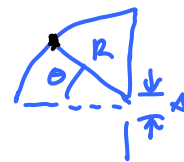
Hemispherical shell



Adapted from Liou, An Introduction to Atmospheric Radiation, 2e

Radiation Escaping

- Volume of radiation $(R d\theta)(R \sin \theta d\phi)(c dt)$
 – has energy = $\frac{U}{V} c dt R^2 \sin \theta d\theta d\phi$



- Photons able to escape = $\frac{A \cos \theta}{4\pi R^2}$

– Energy escaping = $\frac{A \cos \theta U}{4\pi R^2 V} c dt R^2 \sin \theta d\theta d\phi$

- Total energy escaping

$$= \int_0^{2\pi} d\phi \int_0^{\pi/2} d\theta \frac{A \cos \theta}{4\pi} \frac{U}{V} c dt \sin \theta$$

$$= \frac{A}{4} \cdot \frac{U}{V} \cdot c dt$$

power per unit area
 $= \frac{c}{4} \cdot \frac{U}{V}$

Stefan's Law

- Power per unit area = $\frac{c U}{4 V}$
- Knowing total energy, can calculate $\frac{U}{V} = \frac{8\pi^5 (kT)^4}{15 (hc)^3}$
- Power per unit area = $\frac{2\pi^5 (kT)^4}{15 h^3 c^2} = \sigma T^4$
 where $\sigma = \frac{2\pi^5 k^4}{15 h^3 c^2} = 5.67 \times 10^{-8} \frac{\text{W}}{\text{m}^2 \text{K}^4}$
 \uparrow Stefan-Boltzmann constant

Radiating Objects

- For perfect blackbodies, emit and absorb all frequencies
 - Emissivity and absorptivity are both 1
 - If was perfect reflector, emissivity would be 0
- For “greybodies”,
 - Emissivity (ϵ) is less than one and often depends on frequency
 - Power / unit area = $\epsilon \sigma T^4$
- Examples of uses:
 - Infrared thermometry
 - Colour temperature



Colour Temperature

- When buying compact fluorescent light bulbs, they usually quote a "colour temperature" to describe the range of colours

- This value describes the peak in the lamp spectrum by giving an equivalent blackbody temperature

~ 2000 K: Incandescent bulbs

~ 3000 K: Halogen bulbs

~ 5000-6000 K: Summer sun
from Natural Resources Canada

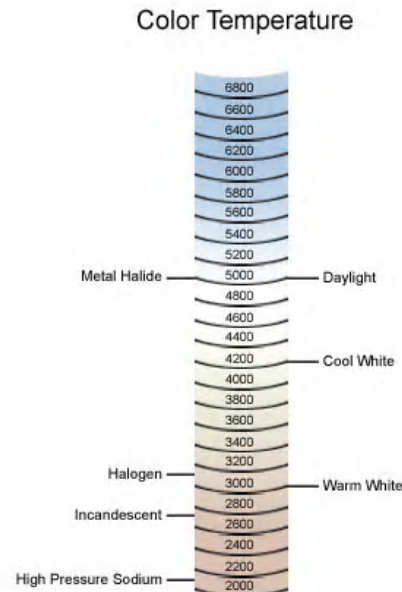


Chart from: <http://www.lightbulbsdirect.com/page/001/CTGY/ColorTemp>

Radiation from Sun and Earth

Sun can be approximated as a blackbody at ~5800 K

- Solar constant 1370 W/m^2 ← radiation at top of atmosphere

radiation from Sun intercepted by Earth → $\text{solar constant} \cdot \pi R^2$

radiation emitted by Earth in all directions → $4\pi R^2 \cdot \sigma T^4$

equate

$$1370 \text{ W/m}^2 \cdot \pi R^2 = 4\pi R^2 \cdot \sigma T^4$$

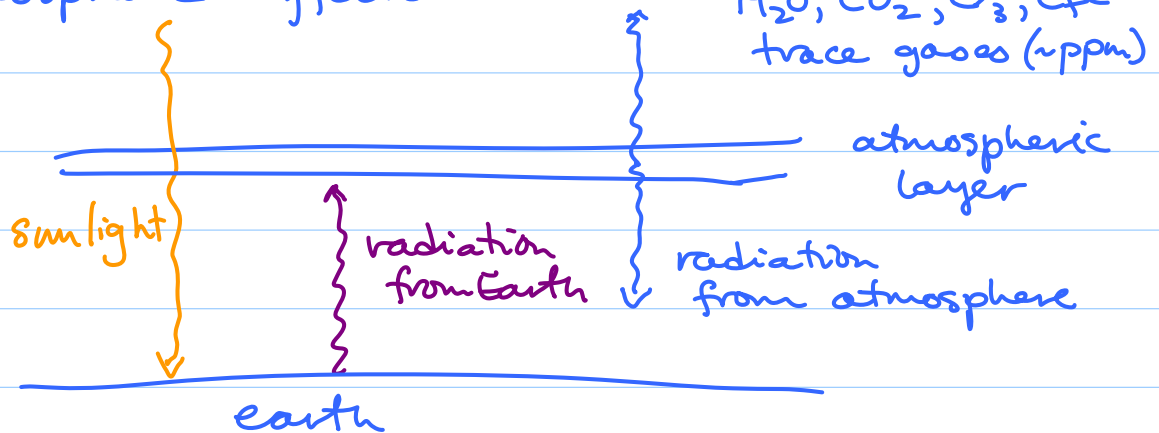
$$T = 279 \text{ K}$$

Some of Sun's radiation is reflected by Earth → 30% albedo

$$1370(1-0.3)\pi R^2 = 4\pi R^2 \cdot \sigma T^4$$

$$T = 255 \text{ K}$$

atmospheric effects



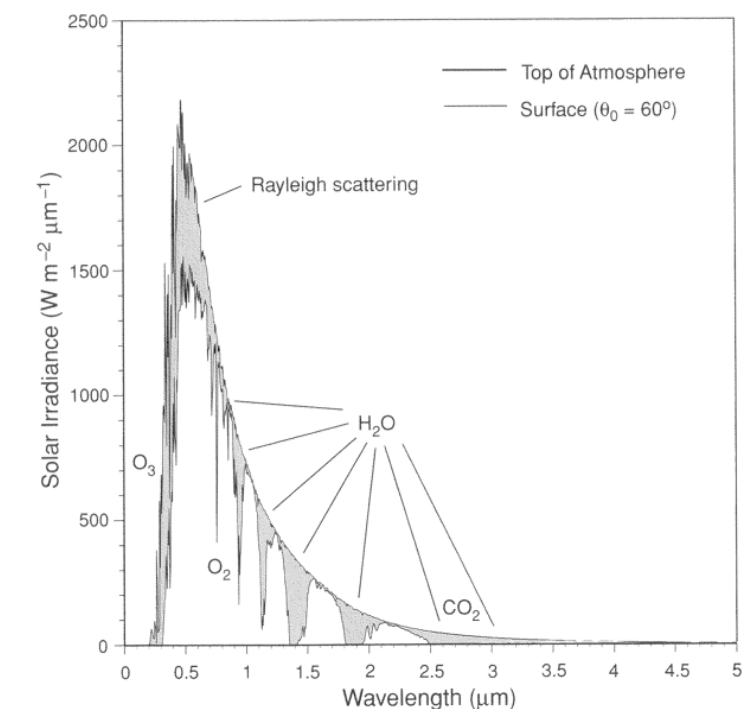
→ if radiation from atmosphere doubles
radiation absorbed by Earth

$$2(1370(1-0.3))\pi R^2 = 4\pi R^2 \cdot \sigma T^4$$

$$T = 303 \text{ K}$$

bit higher than average but shows impact

Absorption by the Atmosphere



← E

Sequence of Spectra of Atmosphere

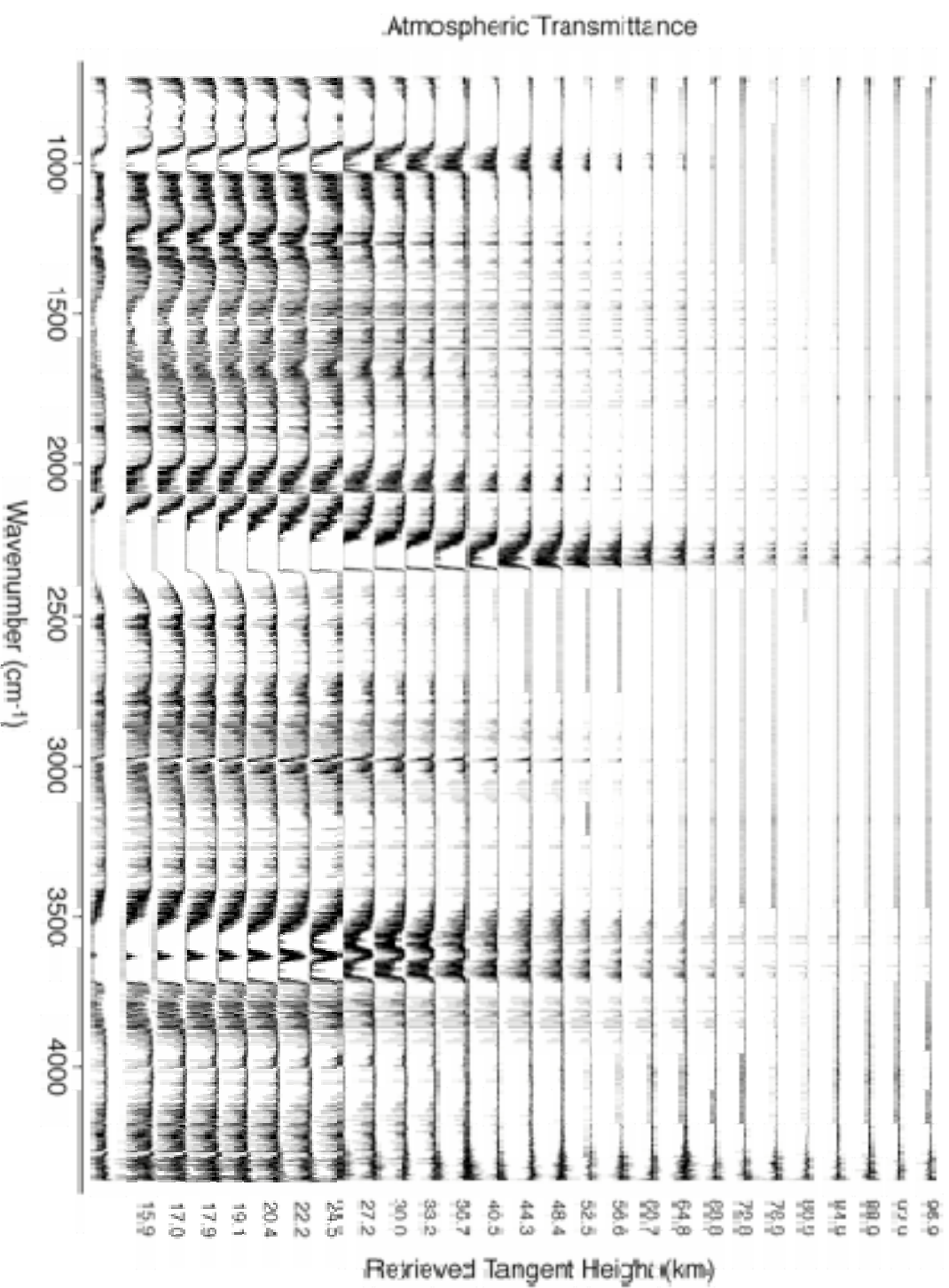
Sunset 2245

Jan. 12, 2004

9:50:23UTC

Lat: 67°S

Lon: 168°W



Cosmic Background Radiation

